

1st International High-Pressure Flame Chemistry Workshop

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# Towards a predictive combustion chemistry model – Uncertainty propagation and minimization

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# **Model Hierarchy**



# **Reaction Model Development**

### **The Current Approach**



### **Kinetic Rate Parameter Uncertainties**

 $H + O_2 \leftrightarrow OH + O (R1)$ 



- Uncertainty factor ~1.25
- Logarithmic sensitivity coefficient
   = 0.24 (ethylene-air, f = 1, p = 1 atm)

- •±5% (±4 cm/s) uncertainty in predicted flame speed due to R1 alone
- •Key question: How do we propagate uncertainties in rate constants in combustion simulations?

### Uncertainty Uncertainty Uncertainty Uncertainty Uncertainty Uncertainty



Burke, et al. (2010)

### Uncertainty Uncertainty Uncertainty Uncertainty Uncertainty Uncertainty



fundamental combustion expt.

# **MUM-PCE**

#### • Method of Uncertainty Minimization – Polynomial Chaos Expansions

• Mathematical foundation and numerical methods: Sheen & Wang "Kinetic uncertainty quantification and minimization using polynomial chaos expansions," *Combustion and Flame*, DOI:10.1016/j.combustflame.2011.05.010.



- Model prediction presented as a (2-*s*) band of uncertainty resulting from kinetic parameter uncertainties.
- Model uncertainty may be constrained by experimental data (ignition delay, species-time history, flame speeds etc)

# **MUM-PCE: Methods**

• Stochastic Spectral Expansion: express kinetic parameter x<sub>i</sub> as a polynomial expansion of basis random variables

$$x_{i} = \mathbf{x}_{i}^{(0)} + \sum_{j=1}^{m} \alpha_{ij} \xi_{j} + \sum_{k=1}^{m} \sum_{j=k}^{m} \beta_{ijk} \xi_{j} \xi_{k} + \dots$$

Following N. Wiener (1938), D.B. Xiu, et al. (2002)

 Solution Mapping: use polynomial response surface to express the relation between a combustion response h and x

$$\eta_r(\mathbf{x}) \cong \eta_{r,0} + \sum_{i=1}^N a_{r,i} x_i + \sum_{i=1}^N \sum_{j\geq i}^N b_{r,ij} x_i x_j$$

## **Forward Uncertainty Propagation**

$$\eta_{r}(\mathbf{x}) = \eta_{r,0} + \sum_{i=1}^{n} a_{i}x_{i} + \sum_{i=1}^{n} \sum_{j \ge i}^{n} b_{ij}x_{i}x_{j}$$
Response surface from solution mapping
$$x_{i} = \frac{1}{2}\xi_{i}$$
Spectral representation of uncertainty in x's (mean = 0, s = 0.5, each indep't of others)
$$\eta_{r}(\mathbf{x},\xi) = \eta_{r}(\mathbf{x}^{(0)}) + \sum_{i=1}^{M} \hat{\alpha}_{r,i}\xi_{i} + \sum_{i=1}^{M} \sum_{j=i}^{M} \hat{\beta}_{r,ij}\xi_{i}\xi_{j}$$

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# **Solution Mapping Method**

• Fit a response surface to the model



<sup>10</sup> G.E.P. Box, *et al.* (1978), Frenklach *et al.* (1992), S.G. Davis *et al.* (2004)



- High-pressure data sensitize kinetics of hydrogen oxidation.
- A large number of models outside experimental uncertainty at high pressures.



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   *σ* uncertainty band calculated by MUM-PCE, based on rate parameter uncertainties.
- Models are statistical samples of parameter uncertainties.

11 Sheen & Wang (2011)

Burke, et al. (2010)

### Uncertainty Uncertainty Uncertainty Uncertainty Uncertainty Uncertainty



fundamental combustion expt.

# **MUM-PCE**



fundamental combustion expt.

# **Method of Uncertainty Minimization**

 $\mathbf{x} = \mathbf{x}_0 + \alpha \boldsymbol{\xi}$ 

Chemical model + associated uncertainty

$$\eta_r(\mathbf{x}) \cong \eta_{r,0} + \sum_{i=1}^N a_{r,i} x_i + \sum_{i=1}^N \sum_{j\geq i}^N b_{r,ij} x_i x_j$$

Physics model

$$\eta_{r}(\mathbf{x},\xi) = \eta_{r}(\mathbf{x}^{(0)}) + \sum_{i=1}^{m} \hat{\alpha}_{r,i}\xi_{i} + \sum_{i=1}^{m} \sum_{j=i}^{m} \hat{\beta}_{r,ij}\xi_{i}\xi_{j}$$

*Predictions* + associated uncertainty

$$\Sigma = \left[ \sum_{r=1}^{n} \frac{1}{\left(\sigma_{r}^{\text{obs}}\right)^{2}} \left( \mathbf{b} \mathbf{x}_{0}^{*} \mathbf{x}_{0}^{*T} \mathbf{b} + \mathbf{a} \mathbf{x}_{0}^{*T} \mathbf{b} + \mathbf{b}^{T} \mathbf{x}_{0}^{*} \mathbf{a}^{T} + \mathbf{a} \mathbf{a}^{T} \right) + 4\mathbf{I} \right]^{-1}$$
$$\boldsymbol{\alpha}^{*} = \boldsymbol{\Sigma}^{1/2}$$

 $k_1$ 

 $\Phi(\mathbf{x}_{0}^{*}) = \min_{\mathbf{x}_{0}} \left\{ \sum_{r=1}^{M} \frac{\left[ \eta_{r,0}^{\text{obs}} - \eta_{r}(\mathbf{x}_{0}) \right]^{2}}{\left( \sigma_{r}^{\text{obs}} \right)^{2}} + \sum_{n=1}^{N} \frac{\left( x_{0,n} \right)^{2}}{\left( \sigma_{n} \right)^{2}} \right\}$ 

*k*<sub>2</sub>

- Model uncertainty constraining
- JetSurF 2.0 H2/CO submodel
  - 14 species, 41 reactions

Dataset 1: From Davis, et al. (2005):

	No.	P <sub>0</sub> , P <sub>5</sub> (atm)	Τ <sub>0</sub> , Τ <sub>5</sub> (K)	f
Laminar Flame Speeds	12	1-15	298	1.0-3.0
Ignition Delay Times	13	0.5-33	1000-2600	1.0-6.1
Flow Reactor Profiles	9	1.0-16	915-1040	0.3-1.0
Laminar Flame Profiles	2	0.047	400	1.9

#### Dataset 2:

From Burke <i>, et al.</i> (2010):	No.	P <sub>0</sub> , P <sub>5</sub> (atm)	<i>Τ</i> <sub>0</sub> , <i>Τ</i> <sub>5</sub> (K)	f
Laminar Flame Speeds	18	15-25	298	0.85 <u>-</u> 2.5





<sup>17</sup> Sheen & Wang (2011)

### Dataset 1 Knowledge prior to 2010



+ Burke, *et al.* (2010) Current knowledge



Weak constraint by experiments Strong constraint by experiments





#### JetSurF – A Jet Surrogate Fuel Model

JetSurF is a detailed chemical reaction model for the combustion of jet-fuel surrogate. The model is being developed through a multi-university research collaboration and is funded by the Air Force Office of Scientific Research. Project participants include

F. N. Egolfopoulos, Hai Wang		University of Southern California		
R. K. Hanson, D. F. Davidson, C. T. Bowman, H.		Stanford University		
Pitsch				
C. K. Law		Princeton University		
N. P. Cernansky, D. L. Miller		Drexel University		
W. Tsang		National Institute of Standards and Technology		
R. P. Lindstedt		Imperial College, London		
A. Violi		University of Michigan		
New Release:	JetSurF Version 2.0 – A working model for the combustion of <i>n</i> -alkane up to <i>n</i> - dodecane, cyclohexane, and mono-alkylated cyclohexane up to <i>n</i> -butyl- cyclohexane ( <i>Release Date: September 19, 2010</i> )			
Old Releases:	Pases: JetSurF Version 1.1 – A interim model for the combustion of n-butyl-, n-propyl-, ethyl-, and methyl-cyclohexane and cyclohexane (Release Date: September 15, 2009)			

### JetSurF Validation – Species Concentrations behind reflected shock waves

B. Sirjean, E. Dames, D. A. Sheen, X.-Q. You, C. Sung, A. T. Holley, F. N. Egolfopoulos, H. Wang, S. S. Vasu, D. F. Davidson, R. K. Hanson, H. Pitsch, C. T. Bowman, A. Kelley, C. K. Law, W. Tsang, N. P. Cernansky, D. L. Miller, A. Violi, R. P. Lindstedt, A high-temperature chemical kinetic model of n-alkane oxidation, JetSurF version 1.0, September 15, 2009 (http://melchior.usc.edu/JetSurF/Version1\_0/Index.html).



Plot stolen from Ron Hanson. Solid line: experiments; dashed line: JetSurF

### **Prediction Uncertainties in As-Compiled Model**



Good nominal prediction with significant uncertainty!

## **Chemistry Model & Experimental Targets**

- Modified JetSurF 1.0
  - 196 species, 1478 reactions

	No.	P <sub>0</sub> , P <sub>5</sub> (atm)	<i>Т<sub>о</sub>, Т<sub>5</sub></i> (К)	f
Laminar Flame Speeds	4	1	353	0.8-1.4
Ignition Delay Times	11	1-4	1000-2600	0.5-2

	No.	P <sub>5</sub> (atm)	Т <sub>5</sub> (К)	f
OH, H <sub>2</sub> O, CO <sub>2</sub> , C <sub>2</sub> H <sub>4</sub> , CH <sub>3</sub> Species Profiles	11	1.6-2.4	1365-1545 K	1

### **Predictions of As-Compiled and Uncertainty-Minimized Models**



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### **Effect on Flame Speed Predictions**



Considering no experiments

#### Model constrained by species profiles

# Model constrained by species profiles + flame speeds

### **Effect on Flame Speed Predictions**



# What did uncertainty minimization do?



### Model constrained by species profiles

H+O<sub>2</sub>↔O+OH HCO+H ↔CO+H, HCO+M↔CO+H+M CH+H<sub>2</sub>↔CH<sub>2</sub>+H CH,+O,↔HCO+OH CH,+O,↔CO,+2H CH<sub>3</sub>+H(+M)↔CH<sub>4</sub>(+M) CH<sub>2</sub>+O↔CH<sub>2</sub>O+H CH3+OH↔CH2,+H2O 2CH<sub>e</sub>(+M)↔C<sub>p</sub>H<sub>6</sub>(+M) G<sub>2</sub>H<sub>2</sub>+H↔G<sub>2</sub>H<sub>2</sub>+H<sub>2</sub> C,H3±O,↔CH°CHQCHO+Q G<sub>0</sub>H<sub>4</sub>+O↔CH<sub>6</sub>+HCO C<sub>2</sub>H<sub>4</sub>+OH↔C<sub>2</sub>H<sub>3</sub>+H<sub>2</sub>O aC<sub>2</sub>H<sub>a</sub>+H(+M)↔G<sub>2</sub>H<sub>a</sub>(+M) Č<sub>s</sub>H<sub>6</sub>+H⇔aC<sub>s</sub>H<sub>5</sub>+H₂ PXG<sub>5</sub>H<sub>11</sub>+G<sub>2</sub>H<sub>5</sub>↔NG<sub>7</sub>H<sub>16</sub> pG<sub>4</sub>H<sub>9</sub>+nG<sub>3</sub>H<sub>7</sub>↔NG<sub>7</sub>H<sub>16</sub> NG7H16+H↔PXG7H15+H2 NG<sub>7</sub>H<sub>16</sub>+H↔SXG<sub>7</sub>H<sub>15</sub>+H<sub>2</sub>

 $\begin{array}{c} \mathsf{H} + \mathsf{O}_{2} \leftarrow \mathsf{O} + \mathsf{O} + \mathsf{O} + \mathsf{H} \\ \mathsf{H} \mathsf{CO} + \mathsf{H} \leftarrow \mathsf{O} + \mathsf{H} \\ \mathsf{H} \mathsf{CO} + \mathsf{H} \leftarrow \mathsf{CO} + \mathsf{H} \\ \mathsf{H} \mathsf{CO} + \mathsf{H} \mathsf{C} \mathsf{O} + \mathsf{H} \\ \mathsf{CH} + \mathsf{H}_{2} \leftarrow \mathsf{CO}_{1} + \mathsf{H} \\ \mathsf{CH}_{3} + \mathsf{O}_{2} \leftarrow \mathsf{H} \mathsf{O} + \mathsf{O}_{1} + \mathsf{H} \\ \mathsf{CH}_{3} + \mathsf{H} (\mathsf{H} \mathsf{M}) \mathsf{C} + \mathsf{CO}_{1} + \mathsf{C} \mathsf{H} \\ \mathsf{CH}_{3} + \mathsf{O}_{2} \leftarrow \mathsf{CO}_{2} + \mathsf{C} \mathsf{H} \\ \mathsf{CH}_{3} + \mathsf{O}_{2} \leftarrow \mathsf{O}_{2} + \mathsf{C} \mathsf{H} \\ \mathsf{CH}_{3} + \mathsf{O}_{2} \leftarrow \mathsf{O}_{2} + \mathsf{C} \mathsf{H} \\ \mathsf{CH}_{4} + \mathsf{O}_{2} \leftarrow \mathsf{O}_{2} + \mathsf{C} \mathsf{H} \\ \mathsf{CH}_{4} + \mathsf{O}_{2} \leftarrow \mathsf{O}_{2} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{3} + \mathsf{H} \leftarrow \mathsf{O}_{2} \mathsf{C} \mathsf{C} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{4} + \mathsf{O}_{2} \leftarrow \mathsf{O}_{2} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{4} + \mathsf{O}_{2} \leftarrow \mathsf{O}_{2} \mathsf{H}_{2} + \mathsf{H} \\ \mathsf{C}_{2} \mathsf{C}_{2} \mathsf{H}_{1} + \mathsf{H} \\ \mathsf{C}_{2} \mathsf{H}_{2} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{3} + \mathsf{H} \leftarrow \mathsf{O}_{2} \mathsf{C}_{2} \mathsf{H}_{2} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{3} + \mathsf{H} \\ \mathsf{C}_{2} \mathsf{H}_{2} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{3} + \mathsf{H} \\ \mathsf{C}_{2} \mathsf{C}_{2} \mathsf{H}_{1} + \mathsf{H} \\ \mathsf{C}_{2} \mathsf{C}_{2} \mathsf{H}_{1} \\ \mathsf{H}_{2} + \mathsf{C}_{2} \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{3} + \mathsf{H} \\ \mathsf{C}_{2} \mathsf{C}_{2} \mathsf{H}_{1} \\ \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} + \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{H}_{2} \\ \mathsf{C}_{2} \mathsf{H}_{2} \\ \mathsf{H}$ 

CH<sub>3</sub>, CH<sub>2</sub>, secondary chain branching, fuel breakup

#### Model constrained by flame speeds



H chain branching

## What did uncertainty minimization do?



# "Our" Approach



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