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# **Further Studies on Spontaneous Ignition of Compressed Hydrogen Releases into Air**

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> Use of hydrogen as an industrial chemical in the manufacture of transportation fuels for upgrading of petroleum and non-petroleum feed stocks is expected to increase as we search for energy solutions in a carbon constrained world.

> Use of hydrogen as an energy carrier in stationary and transportation applications requires careful consideration of safety related matters and continuing evolution of design standards.

> Understanding the sources of and conditions that can lead to ignition is important to advancing safe operating criteria and application of design standards.

> One such ignition mechanism that is nearly unique to hydrogen is the release of compressed hydrogen into air.1

#### Spontaneous Ignition from Sudden Compressed Hydrogen **Release into Air through Pipes§**





Video at http://www.princeton.edu/~combust/research/h2\_safety/ Gas release forms shock that heats air in contact

with expanding H2. Mixing regions near the contact surfaces must achieve temperatures sufficient for chemical ignition.

Sufficiently heated regions must also achieve minimum critical volume of flammable H2/Air mixture ratios to sustain transition from local ignition to inflammation.

Turbulent burning characteristics control transition to sustained turbulent jet flame.

In this configuration, transient boundary layer phenomena within the pipe are principally responsible for producing the required mixing.

- Partial shock reflections from reductions in pipe diameter can result in required >> spontaneous release heating at pressures as low as 300 psia.
- > In all cases, a minimum pipe length is required to achieve sufficient mixing for spontaneous ignition to occur, even if H2 release pressure is sufficiently high for heating requirement.
- >> Both initial compressed pressure and downstream geometry can be critical factors in promoting spontaneous ignition.1
- Controlling phenomena interact non-linearly and require more experimental and numerical study to determine geometry effects and scaling.

### Spontaneous Ignition Induced by Compressed Hydrogen **Release Interaction with Downstream Bodies**

> Investigate transient shock interactions with simple two dimensional wedge geometries fully-coupled, transient numerical and simulations with detailed chemical kinetics and diffusive transport to investigate wide range of initial parameters and geometry effects



- Fully Coupled Numerical Modeling (A-SURF (1D, 2D) - (Adaptive Simulation of -2 Unsteady Reacting Flow).
- Methodology is adapted from prior laminar flame research.<sup>3-5</sup> X (cm)

Finite volume, unsteady, compressible N-S equations for multi-species reactive flow:

Reaction term: Strang splitting, VODE solver Diffusion term: Finite difference scheme Convection term: 2nd-order MUSCL-Hancock TVD Scheme

Time advancement: High order explicit TVD Runge-Kutta scheme Dynamically adaptive mesh refinement, (local mesh addition and removal), 8 to 10 levels with finest mesh size (16~4  $\mu$ m) at contact surface.

Detailed chemistry<sup>2</sup>, multi-component transport, and radiation (optical thin & F-SNB-ck models).



- > Lower pressure ratio results in weaker ignition and flame does not propagate into free stream.
- Higher pressure ratio results in higher temperatures, stronger ignition and flame propagates along contact surface.
- Increased wedge angle enhances mixing downstream of wedge shoulder and promotes flame propagation.
- > Reduced wedge angle results in reduced temperatures near wedge surface and delayed ignition.

### **Confirming Experimental Observations**





- Transient shock properties are affected by the pressure boundary rupture characteristics
- Shock structure is not modeled accurately by a thin discontinuity, uni-dimensional representation



Test conditions: 4" pipe length, 1/2" wedge, burst pressure 1715 psig. Refraction of light due to density gradients indicates sustained burning Video at http://www.princeton.edu/~combust/research/h2 safety

- > Blunt body interactions with transient shocks formed by compressed hydrogen release can result in spontaneous ignition.
- Shock must be sufficiently strong and/or focused by the interaction to result in temperatures exceeding local autoignition temperatures.
- Critical mixing at the contact surface must also be induced by the shock/body interaction.

#### References

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Test conditions: 4" pipe length, 1/2" wedge, burst pressure 1715 psig Refraction of light due to density gradients indicates sustained burning after spontaneous ignition event

Combinations of pipe length and wedge geometries downstream of the pipe exit have been investigated

	Test No.	Burst Disk	Configuration	Burst Pressure (Psig)	Result
	1	0.007" Cu	4" pipe	1728	No Ignition
	2	0.007" Cu	12" pipe	1743	Ignition
	3	0.007" Cu	4" pipe, wedge 1/2° downstream	1762	Ignition
	4	0.007" Cu	No pipe <sup>1</sup> , wedge 1/2° downstream	1733	?
	5	0.007" Cu	4" pipe, wedge 1/2° downstream	1715	Ignition
	~1" female pipe threaded region downstream of burst disk				