

Problem 71, Chemistry 301X - 2006

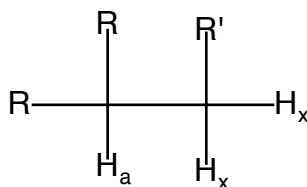
(a) One might expect that a two-spin system of the following kind would show two lines, one for each different hydrogen. But, as a former US president once said, that would be wrong.



Remember that each *different* hydrogen - each hydrogen in a different molecular environment - resonates at a *different* frequency (energy), and that the spin of H<sub>x</sub> (or of H<sub>a</sub>) can have two values, + 1/2 or - 1/2. Now predict the NMR spectrum of the “AX” system shown above.

Get your answer checked!

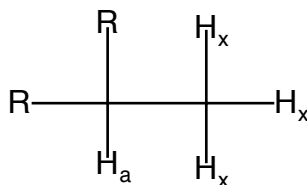
(b) Now predict the spectrum for the following “AX<sub>2</sub>” system:



First catalog the possible spin combinations of two H<sub>x</sub> hydrogens, then predict the spectrum.

Get your answer checked!

(c) Now do the same thing for “AX<sub>3</sub>”



Finally, generalize: how many lines will appear when a hydrogen, H<sub>a</sub>, is flanked by *n* adjacent equivalent hydrogens? Draw a molecule in which an H<sub>a</sub> is flanked by four equivalent H<sub>x</sub>'s.