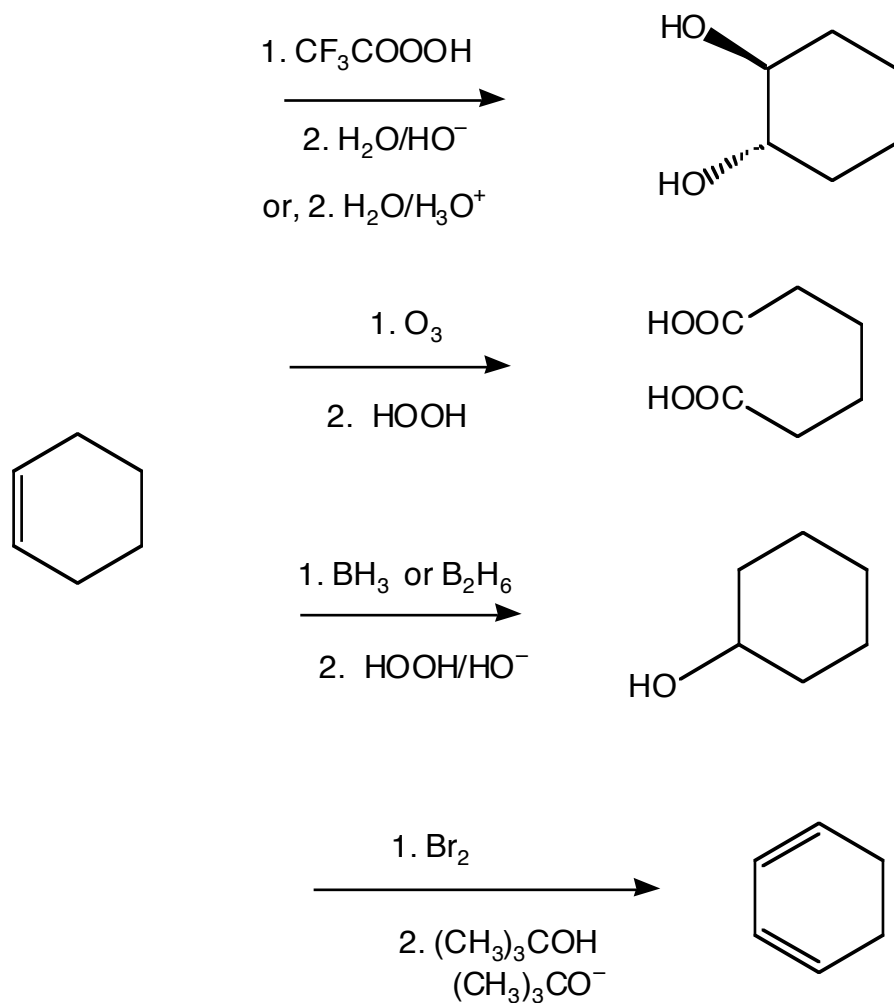


Answers to problem 77, Chemistry 301X - 2006



3b. There are, of course many ways to do each. From the top:

a) real compound will have strong OH stretch at $\sim 3300 - 3500 \text{ cm}^{-1}$

b) real compound will lack the “aldehyde couplet” at 2850 and 2750 cm^{-1}

(c) real compound has no low field signal for the $\text{C}=\text{O}$ carbon in the ^{13}C NMR spectrum ($\sim 200 \text{ ppm}$)

(d) real product has three different carbons (and three different hydrogens) whereas the wrong compound has two different carbons and two different hydrogens.