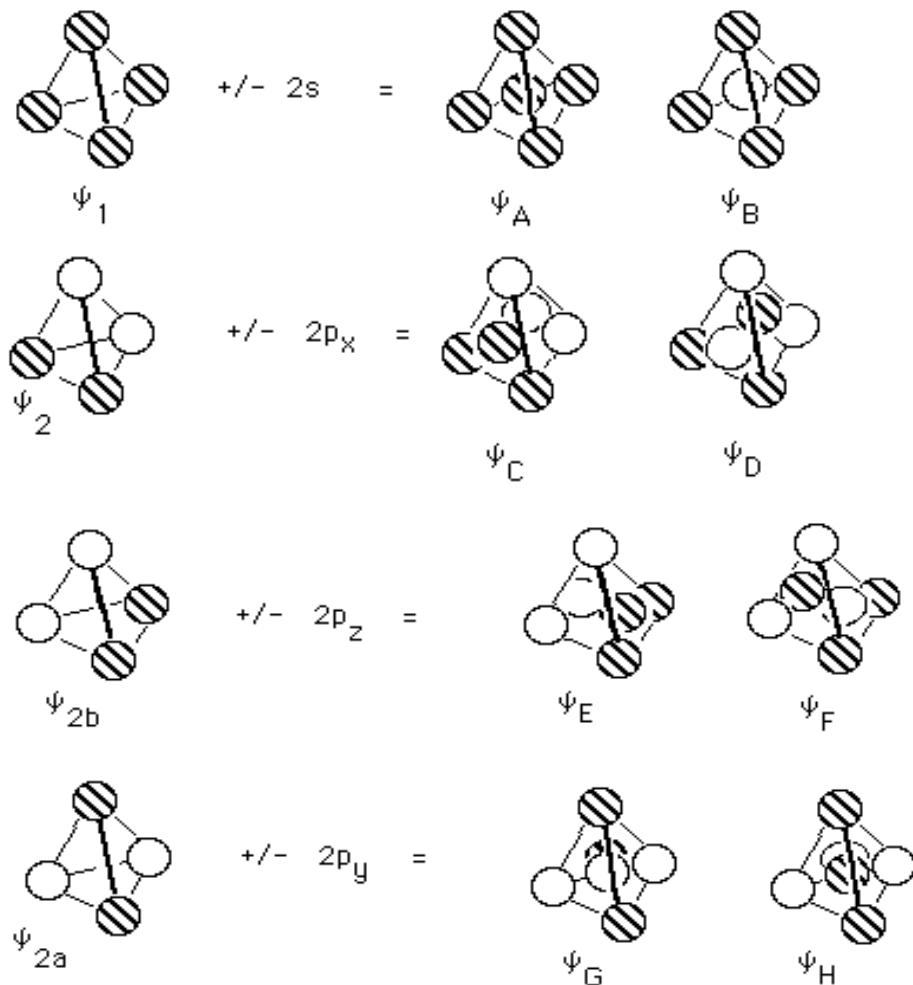


**An answer to ad hoc problem set #1 - constructuion of the molecular orbitals of methane:**

Let the orbitals of tetrahedral  $H_4$  interact with the atomic orbitals of carbon of the proper symmetry. This means  $1 +/- 2s$ ,  $2 +/- 2p_x$ ,  $2a +/- 2p_y$  and  $2b +/- 2p_z$ .



The levels and electronic occupancy for the eight-electron system (4 from C, 4 x 1 from the H's) will be:

<u>D</u>	<u>B</u>	<u>G</u>	—	—	—
<u>C</u>	<u>E</u>	<u>H</u>	↑↑	↑↑	↑↑
	<u>A</u>			↑↑	

