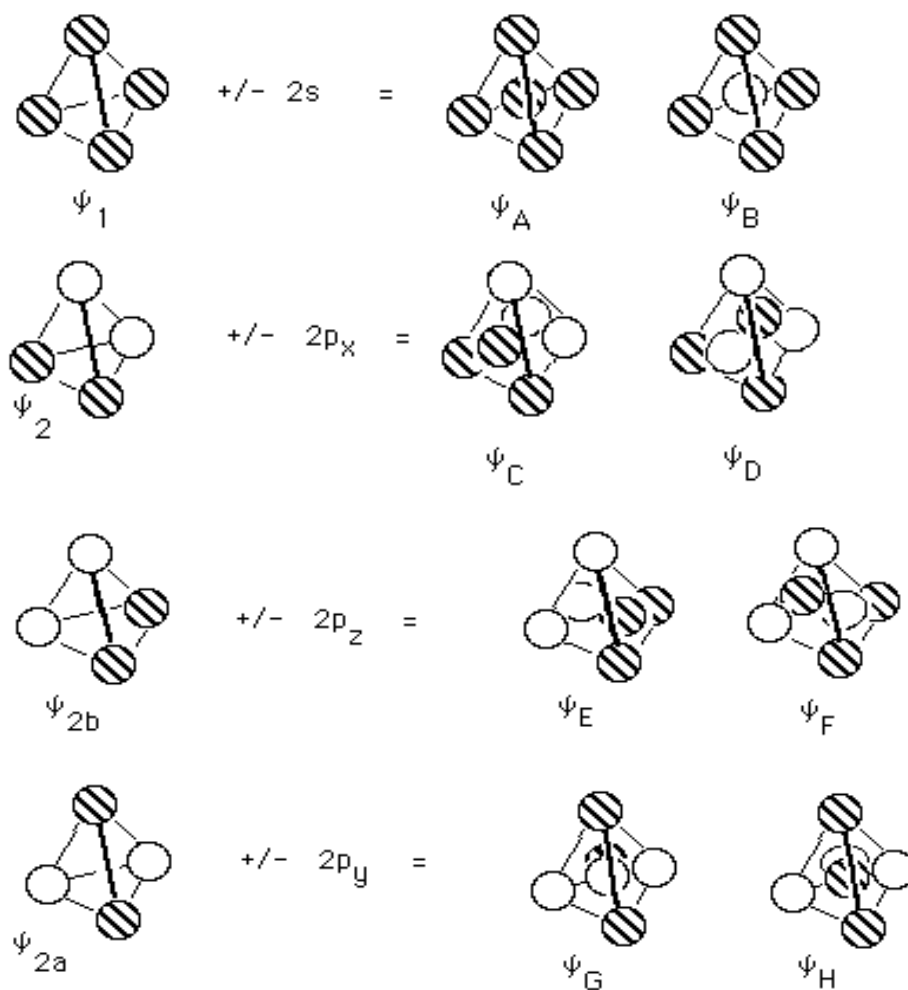


An answer to ad hoc problem set #1 - construction of the molecular orbitals of methane:

Let the orbitals of tetrahedral H_4 interact with the atomic orbitals of carbon of the proper symmetry. This means ψ_1 \pm $2s$, ψ_2 \pm $2p_x$, ψ_{2a} \pm $2p_y$ and ψ_{2b} \pm $2p_z$.



The levels and electronic occupancy for the eight-electron system (4 from C, 4 x 1 from the H's) will be:

