

Investigation of Methane and Syngas Autoignition at High Pressure Conditions

Wenting Sun
School of Aerospace Engineering
Georgia Institute of Technology
Atlanta, GA 30332

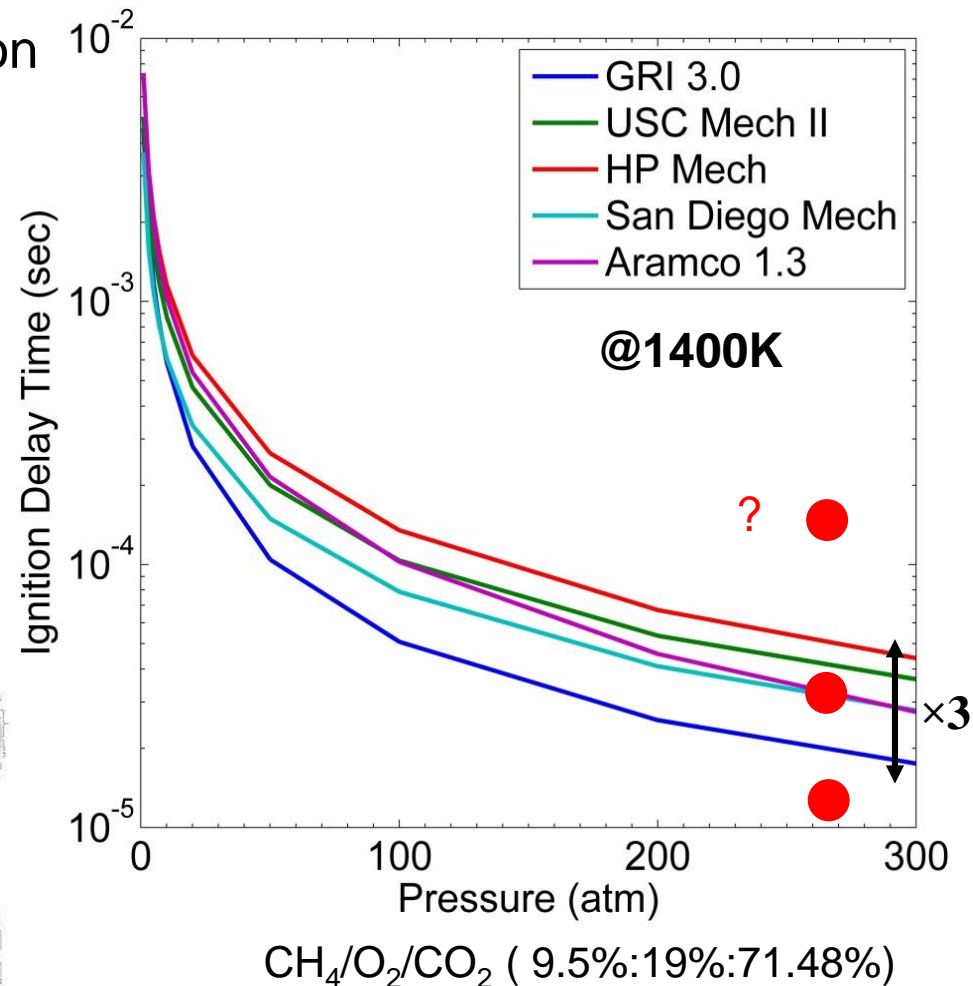
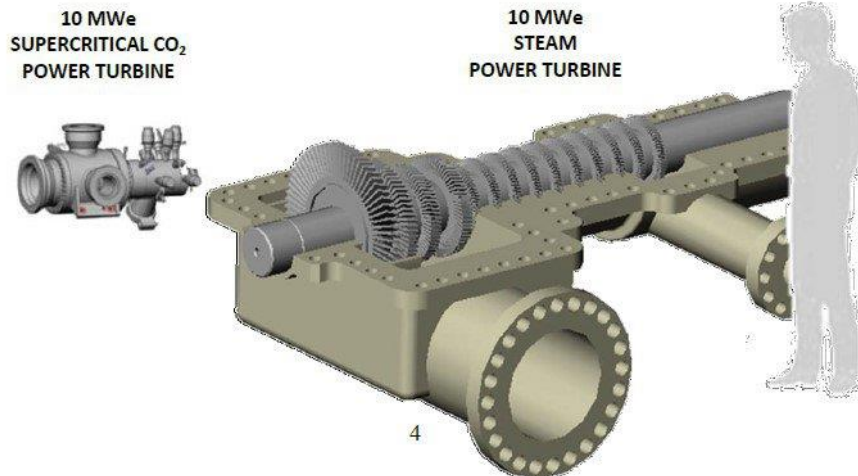


The 2nd International Workshop on Near-limit Flames
Beijing, China

Background

- Technology driven
 - Supercritical CO₂ oxy-combustion
 - high efficiency, ~100% carbon capture
 - Completely new operating condition
 - 100-300 atm
- Bridge the knowledge gap

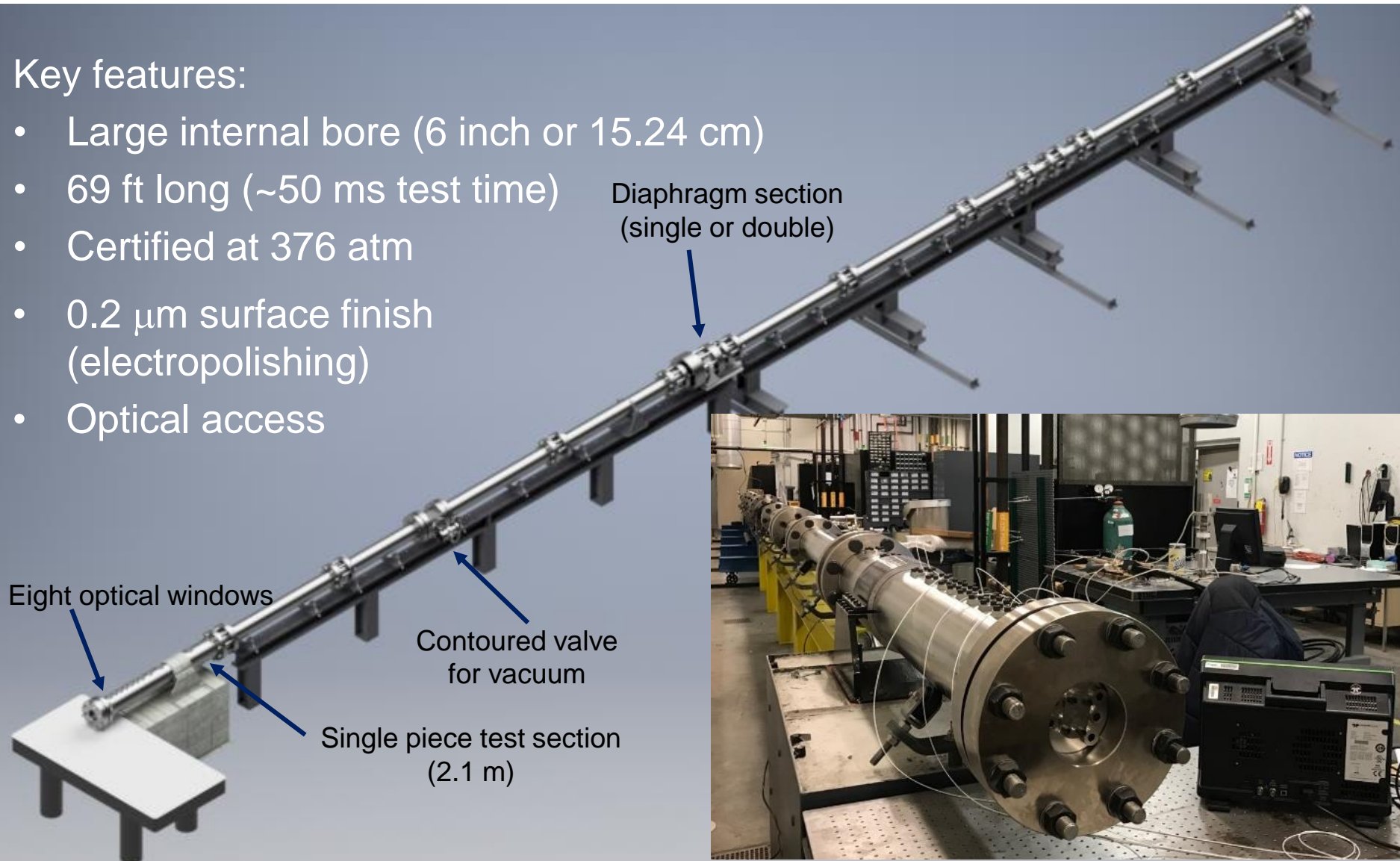
Compact but more efficient system



GT High Pressure Shock Tube

Key features:

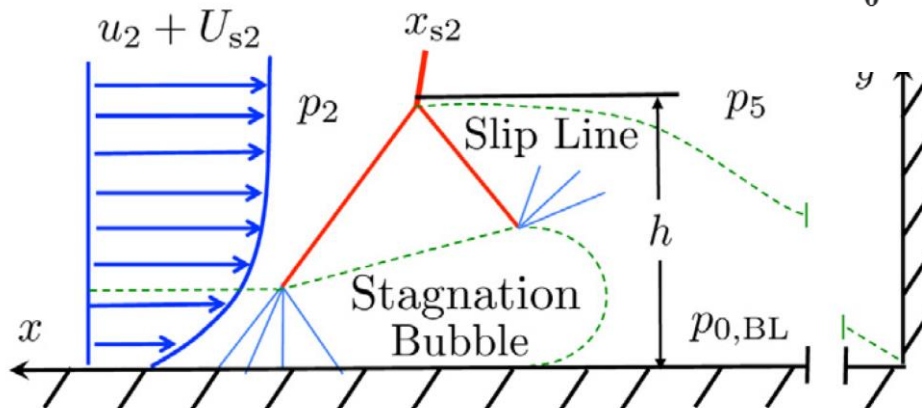
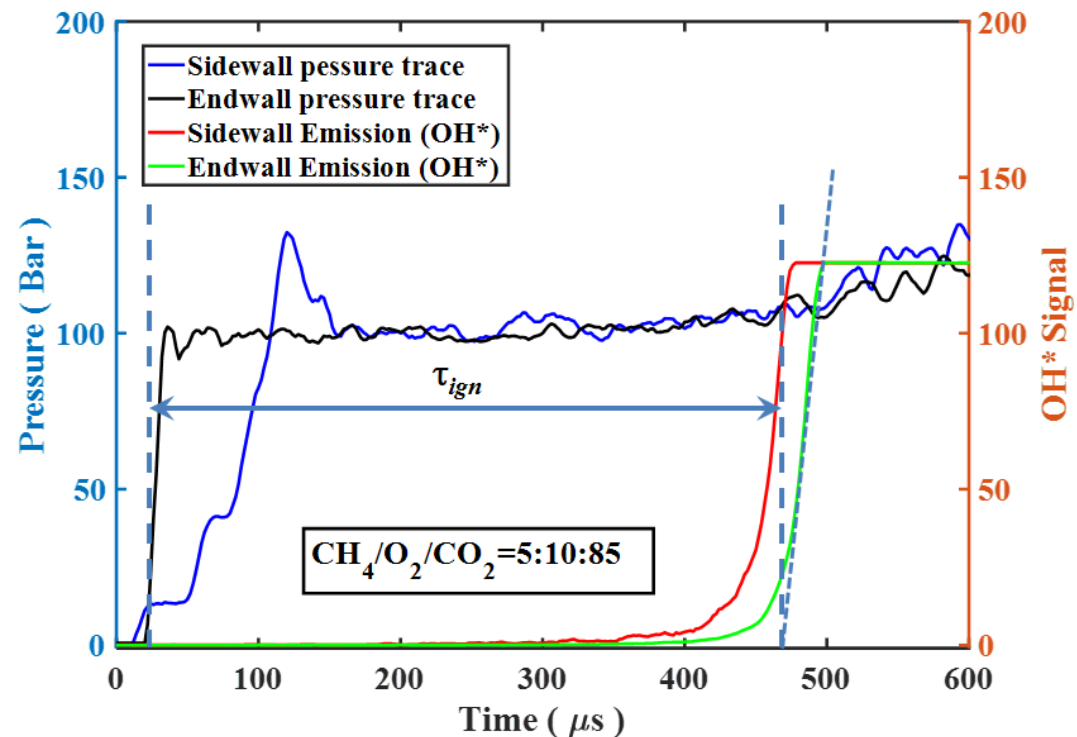
- Large internal bore (6 inch or 15.24 cm)
- 69 ft long (~50 ms test time)
- Certified at 376 atm
- 0.2 μm surface finish (electropolishing)
- Optical access



Challenges of Experiments at sCO₂ Condition

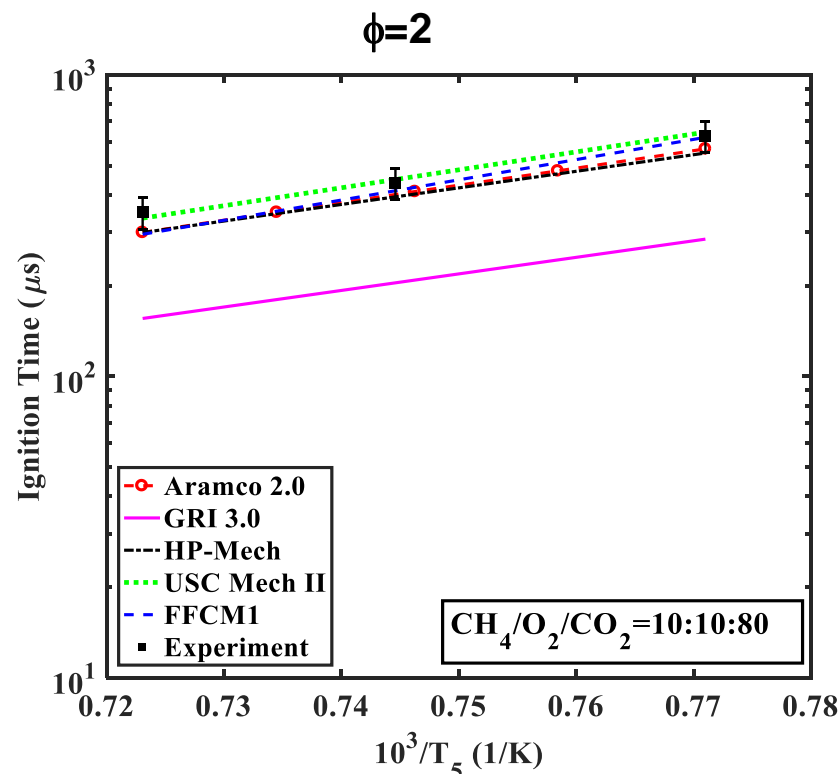
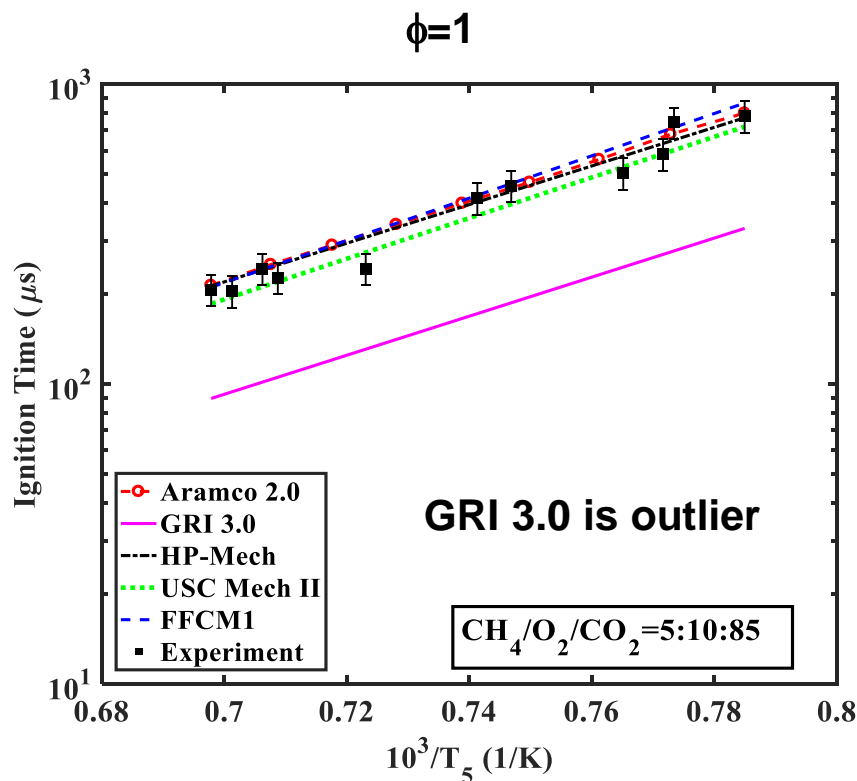


- Large C_p of CO₂
 - Strong shock needed for CO₂
- BL is much thicker with CO₂
 - Non-ideal effect
- ID of shock tube must be large
 - 150 mm
 - High experimental cost



Autoignition Delays at sCO₂ condition

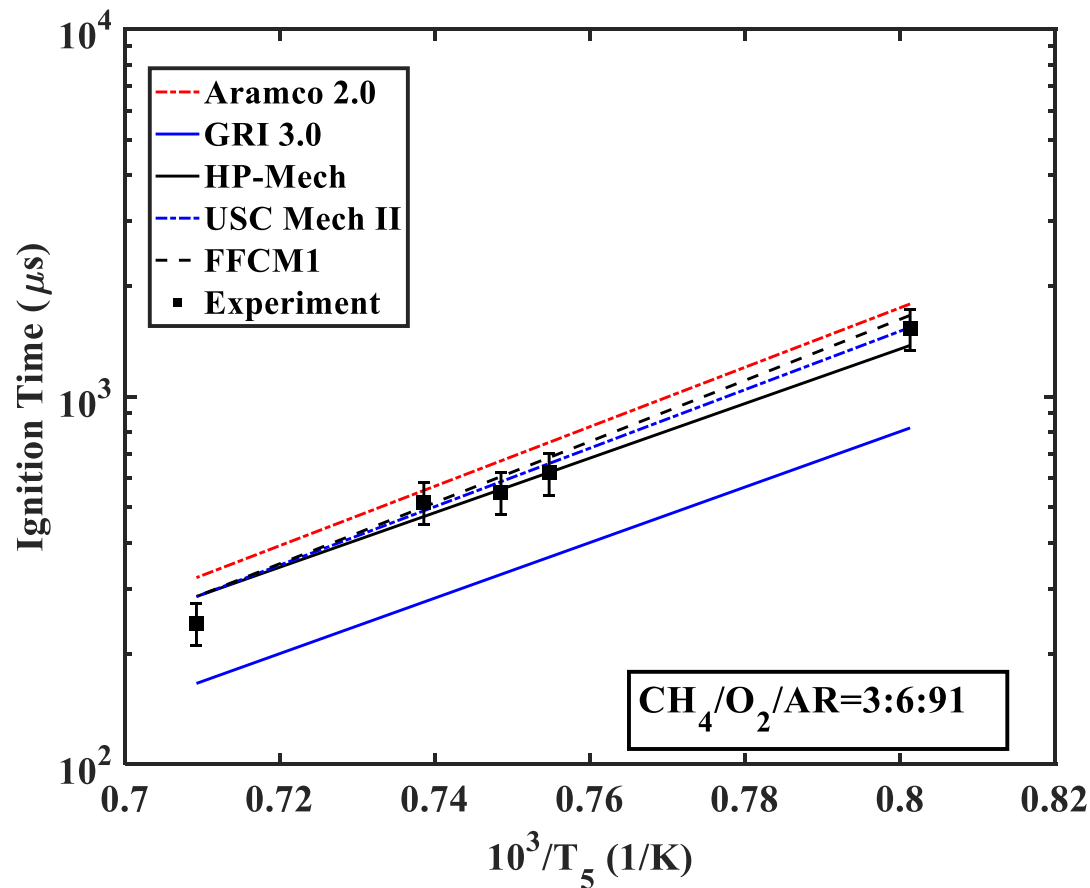
- Pressure: 100±5 bar, and temperature range of 1274 to 1433 K



Simulation results from Aramco 2.0, USC Mech II and HP-Mech are close to each other, however GRI 3.0 predicts a significantly shorter autoignition delay, having approximately a **factor of 3 difference**

Autoignition Delays of CH₄ in Ar

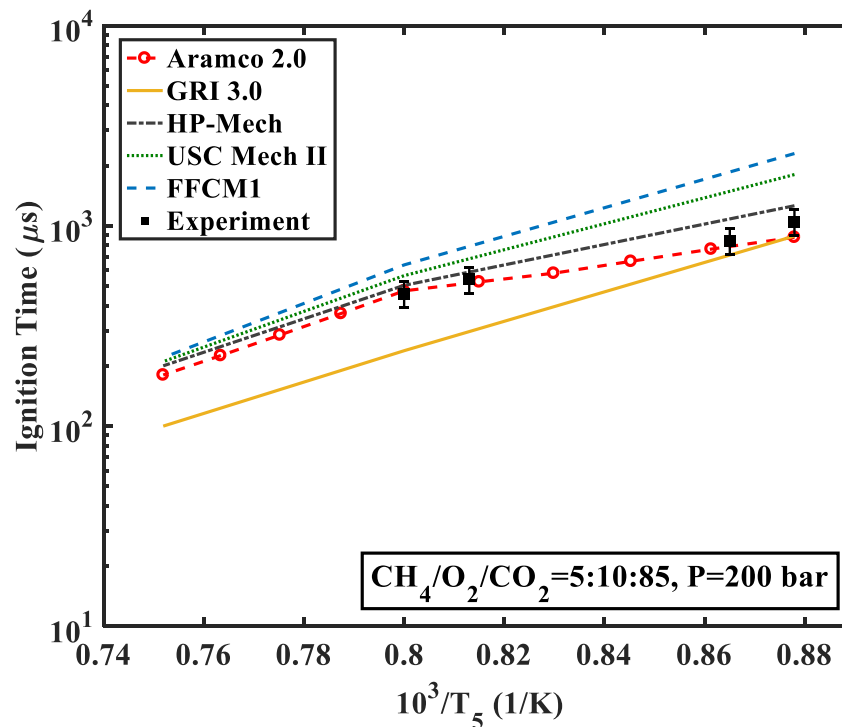
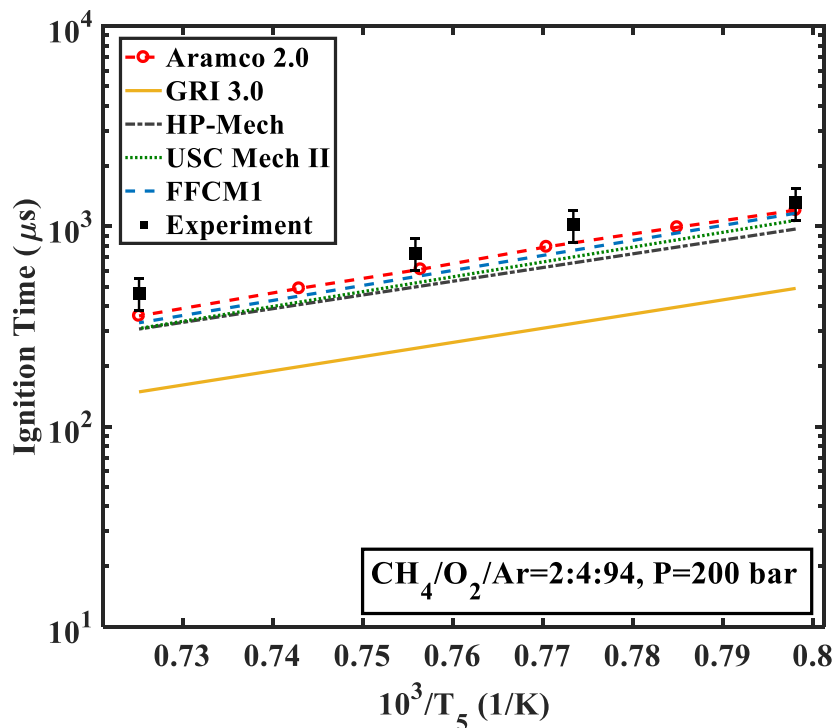
- Pressure: 95±3 bar, and temperature range of 1248 to 1410 K



No chemical effect from diluent is observed

Autoignition Delays of CH₄ in CO₂ and Ar

- Pressure: 200±5 bar, and temperature range of 1137 to 1380 K

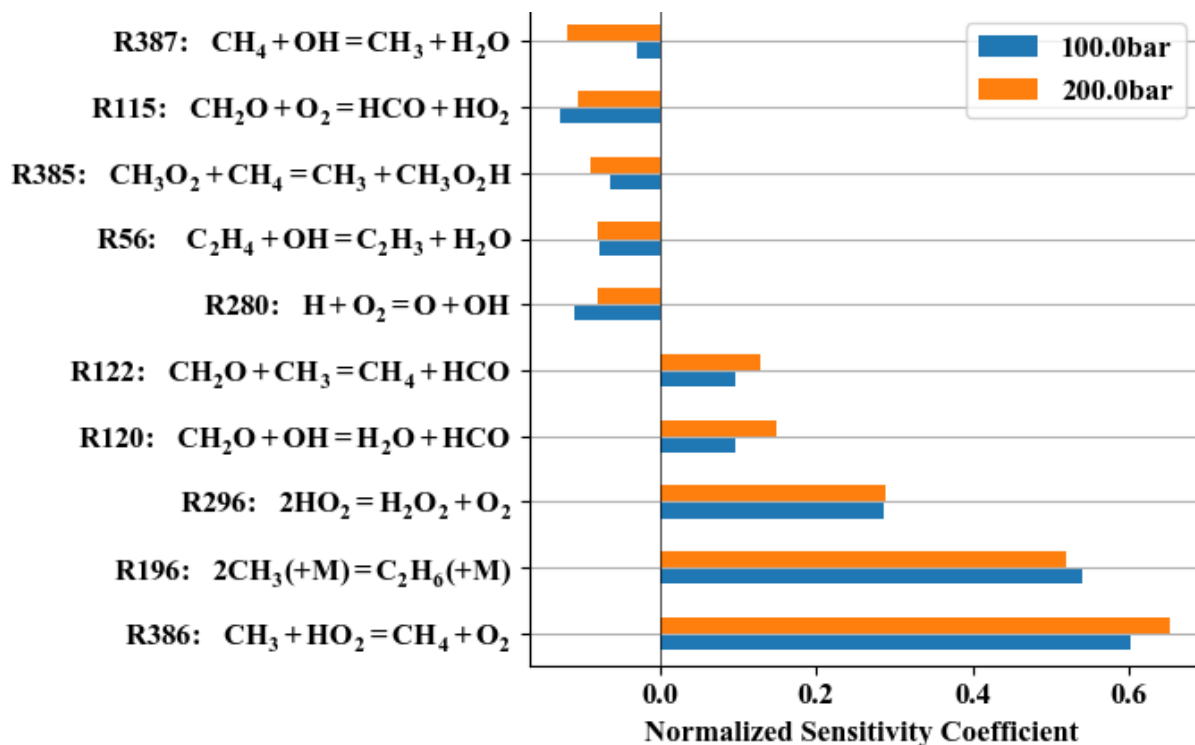


Temperature further distinguishes different kinetic models
 - High T kinetics is much simpler than low T kinetics



Chemical Analysis

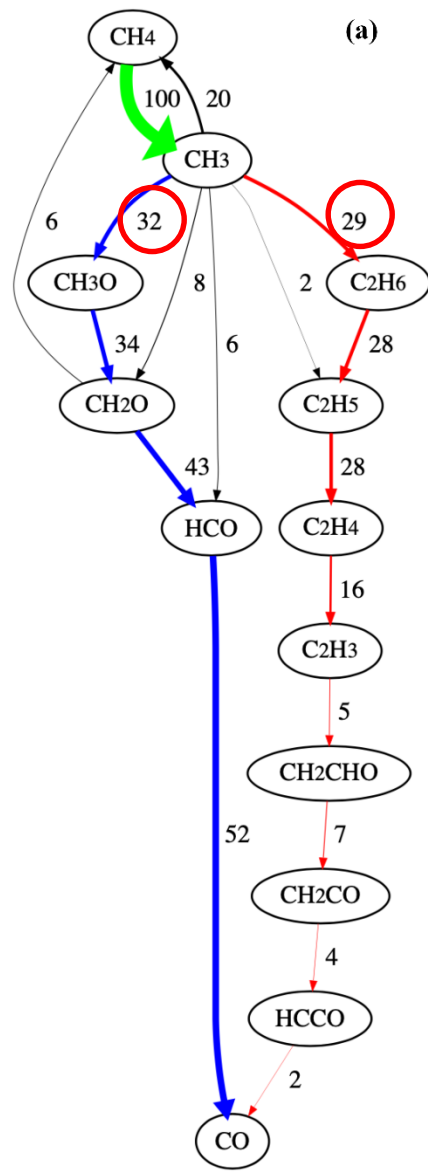
- $\text{CH}_3 + \text{CH}_3 + \text{M} = \text{C}_2\text{H}_6 + \text{M}$, M should matter
 - It is approaching high pressure limit...
 - How about low pressure?
- The third body efficiency of CO_2 was investigated by doubling its value, which has a negligible effect on autoignition from 1-15 atm
 - J.W. Hargis, E.L. Petersen, Energy & Fuels, (29) 2015
 - S. Vasu, D.F. Davidson, R.K. Hanson, Energy & Fuels, (25) 2011



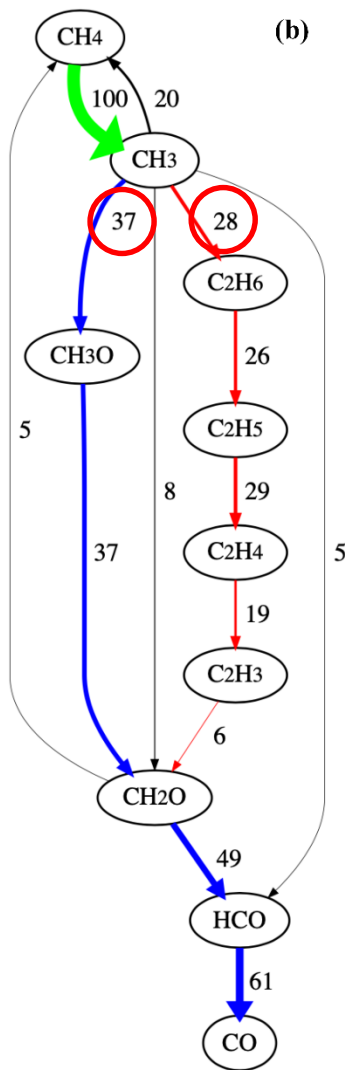
T=1200 K
 $\text{CH}_4/\text{O}_2/\text{CO}_2$ mixture
(5:10:85)

CH₄ Reaction Pathway Analysis

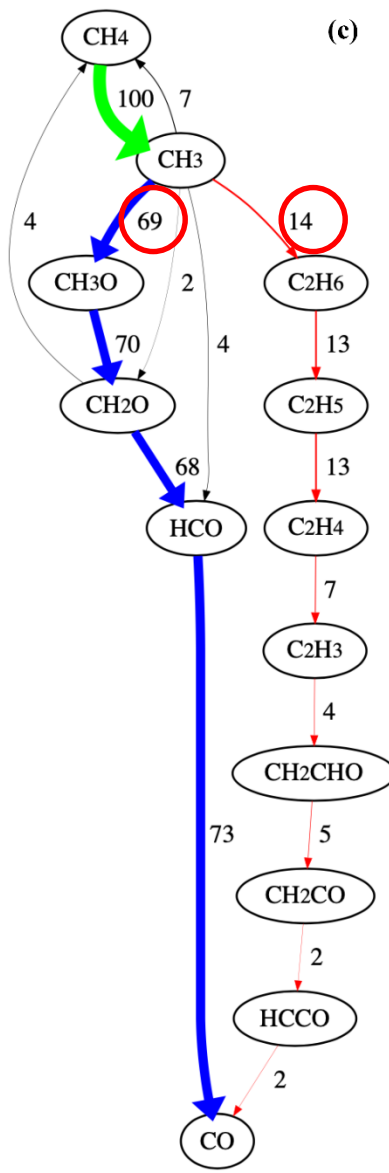
Aramco 2.0



FFCM-1



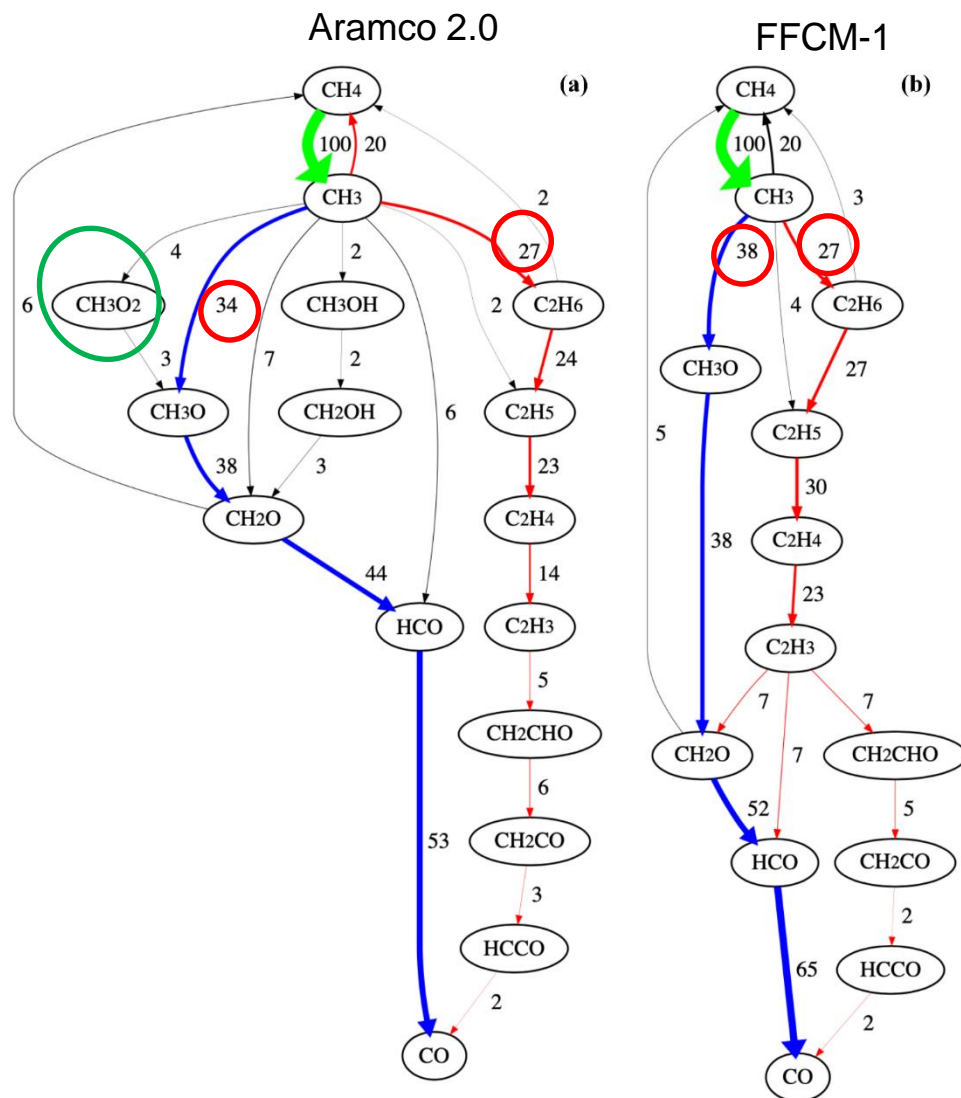
GRI 3.0



P=100 bar, T=1200 K
CH₄/O₂/CO₂ (5:10:85)

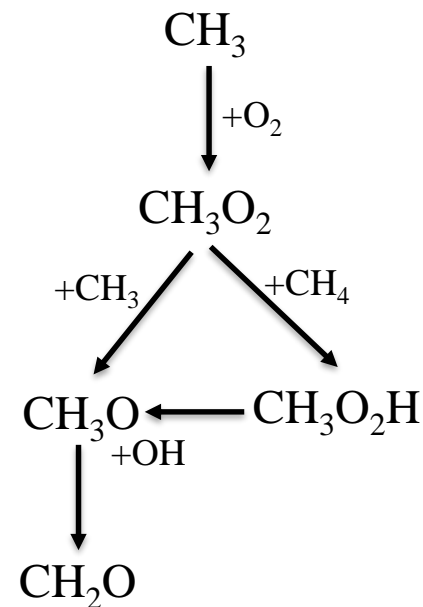
CH₄ branching ratio
dictates ignition

CH₄ Reaction Pathway Analysis



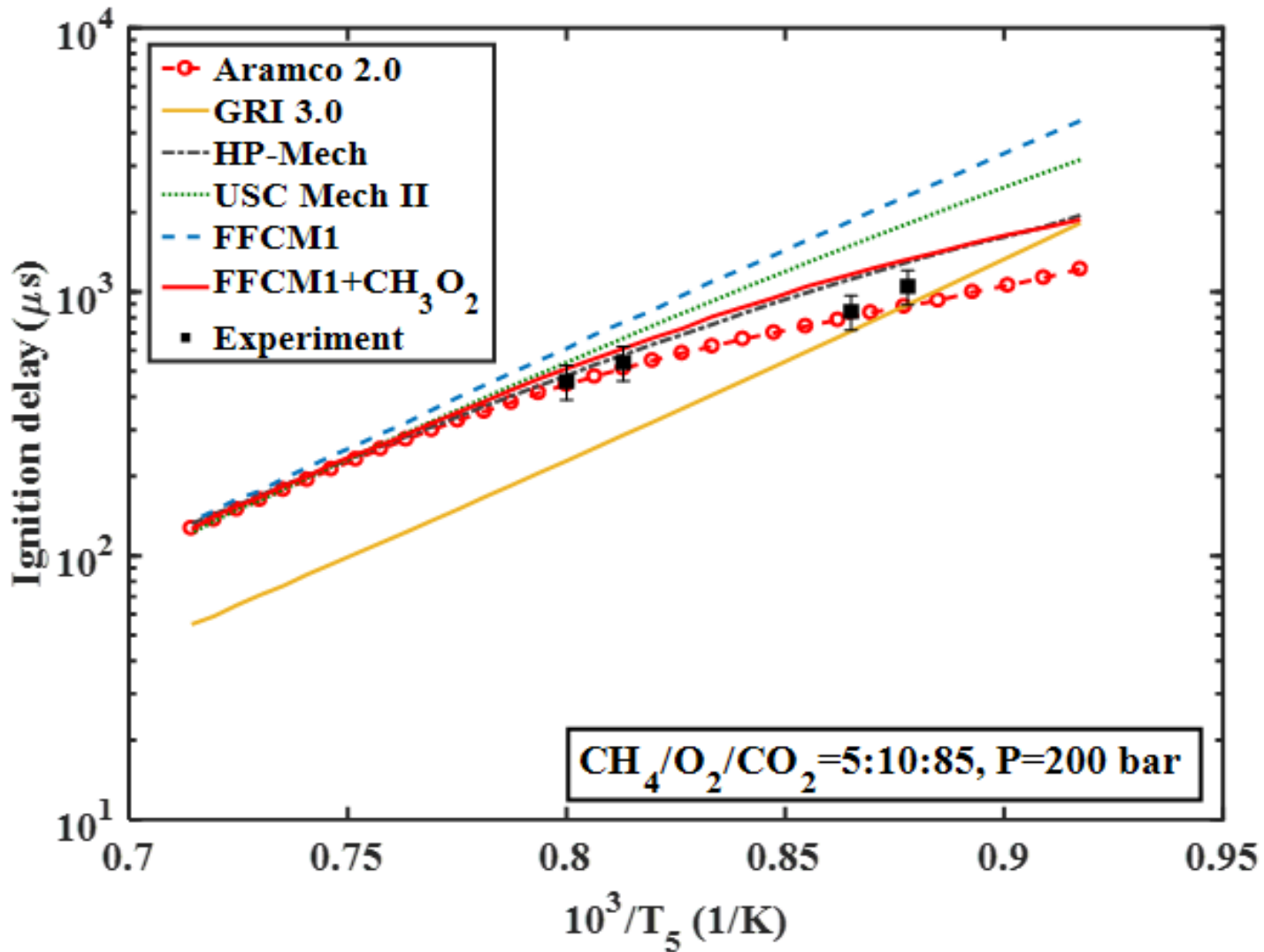
P=200 bar, T=1200 K
CH₄/O₂/CO₂ (5:10:85)

Additional pathway of CH₃ at low temperature



CH₃O₂ is NOT included in FFCM

CH₄ Autoignition Delays





Conclusion

- High pressure CH_4 and syngas autoignition delays were measured at 100 and 200 bar
- CH_3O_2 is important at high pressure and relatively low temperature conditions
- CO_2 has chemical effect on elementary reactions through third body collision efficiency, but the overall effect on ignition is washed out

Thank you! & Questions?



Acknowledgement: DoE UTSR Project