



Transition Metal Catalyzed Cross Coupling Reactions of Unactivated Alkyl Halides

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8/2/2006

Transition Metal-Catalyzed Cross Coupling Reactions

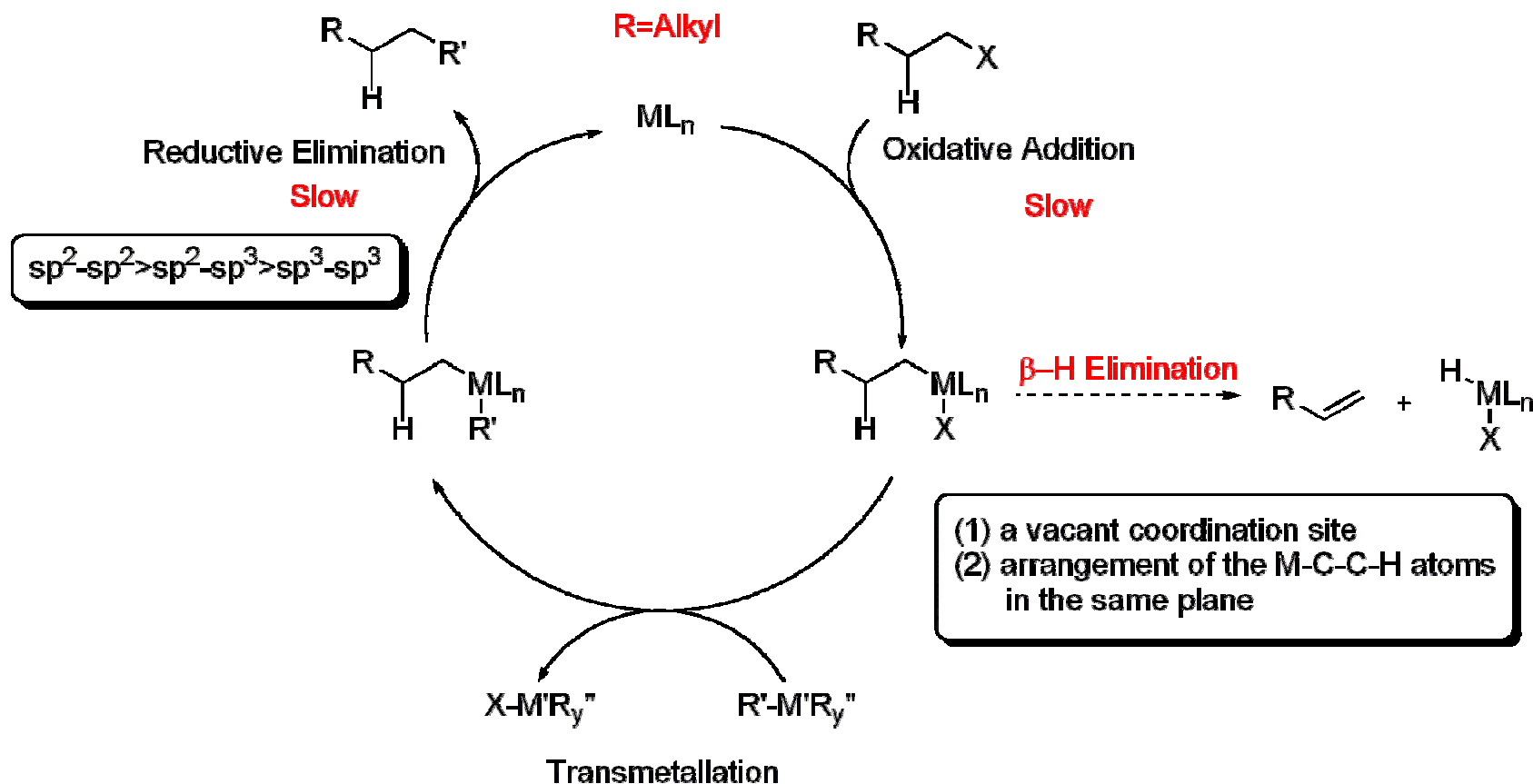


Named Reaction	R-M	R'-X
Kumada-Corriu	R-MgX or R-Li	aryl, alkenyl
Suzuki-Miyaura	R-BR' ₂	aryl, alkenyl, alkyl
Negishi	R-ZnX	aryl, alkenyl, alkynyl, acyl
Stille	R-Sn(alkyl) ₃	aryl, alkynyl, acyl
Heck	alkene	aryl, alkenyl, alkyl (no β hydrogen)

Unactivated alkyl halides: that are not functionalized with activating groups in the α position.

Activated alkyl halides: benzyl, allyl, propargyl, cyclopropyl, α-alkoxy, α-carbonyl alkyl halides

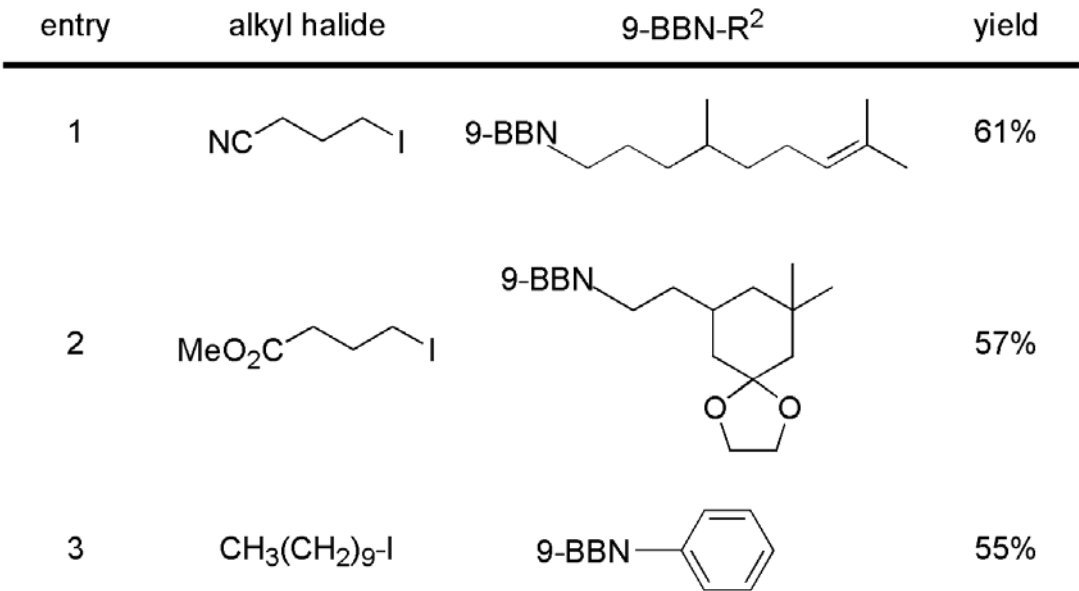
General Mechanism



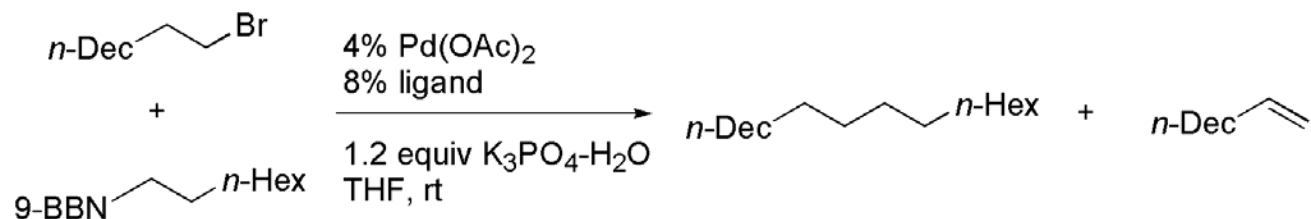
Transition Metal Catalyzed Cross Couplings of Unactivated Alkyl Halides

- ◇ Suppress β -hydride elimination
- ◇ Facilitate oxidative addition
- ◇ Facilitate reductive elimination
- ◇ Utilize new mechanisms

➔ Transition Metal / Ligand / Solvent / Base / Additive



Suzuki Coupling-Alkyl Bromides

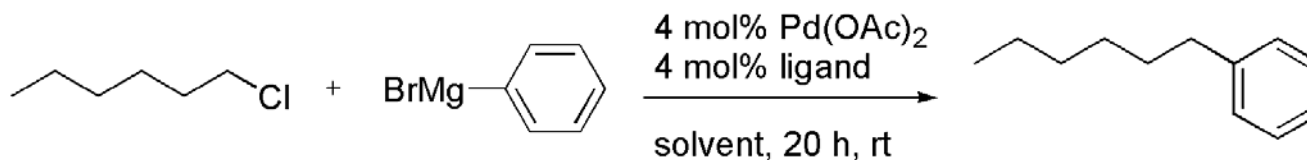


entry	ligand	cone angle (°)	product yield (%)	alkene yield (%)
1	PCy ₃	170	85	<2
2	P(<i>i</i> -Pr) ₃	160	68	6
3	P(<i>t</i> -Bu) ₃	182	<2	21
4	P(<i>n</i> -Bu) ₃	132	9	27
5	dcpe		<2	21
6	PPh ₃		<2	<2

dcpe: 1,2-bis(dicyclohexylphosphino)ethane

Netherton, M. R.; Dai, C.; Neuschütz, K.; Fu, G. C.* *J. Am. Chem. Soc.* **2001**, 123, 10099.

Kumada Coupling (Alkyl Chlorides)

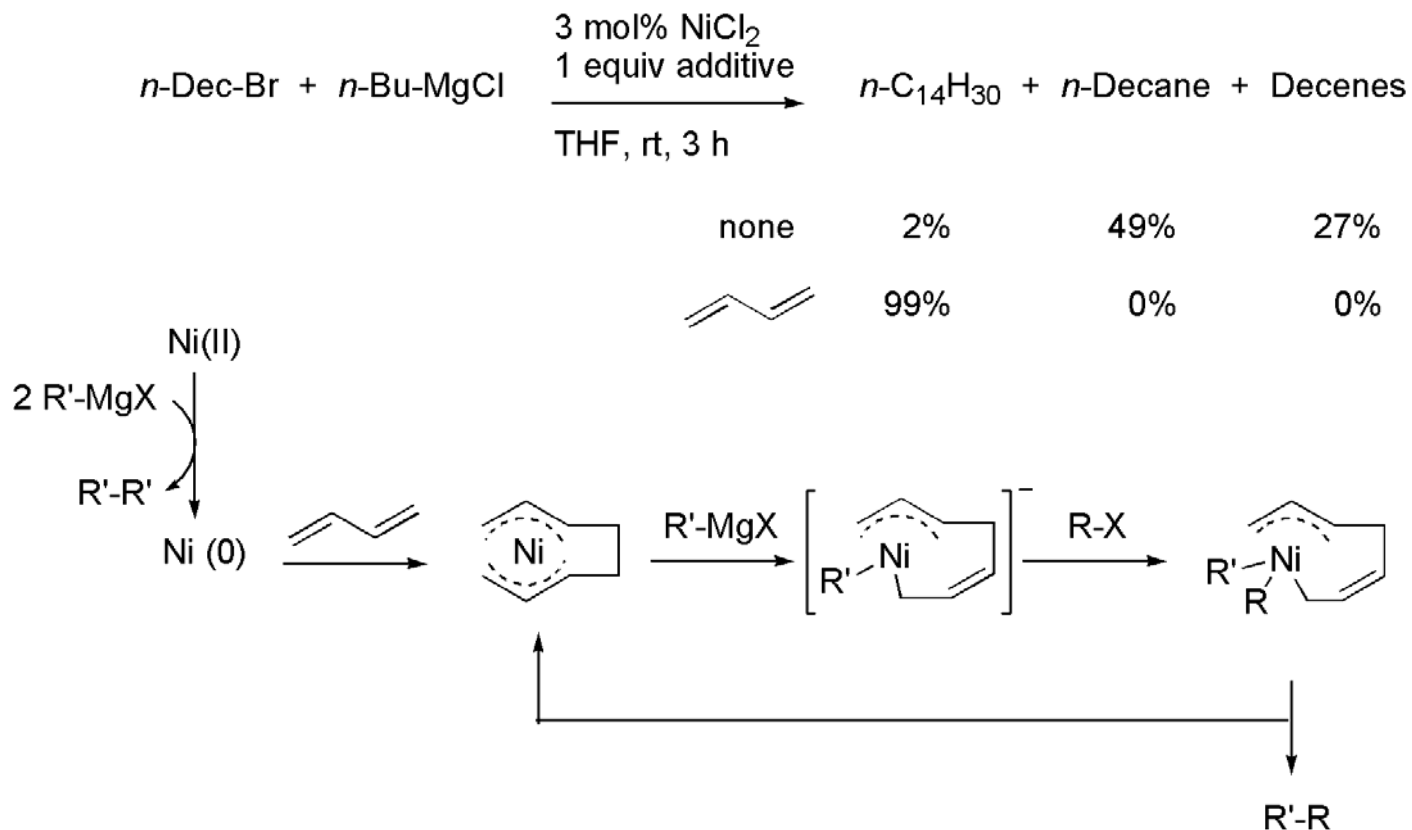


entry	solvent	Ph-MgBr (eq)	Ligand	conversion (%)	yield (%)
1	THF	1.2	PCy ₃	1	1
2	dioxane	1.2	PCy ₃	1	2
3	DMF	1.2	PCy ₃	70	3
4	DMAc	1.5	PCy ₃	95	84
5	NMP	1.5	PCy₃	99	96
6	NMP	1.2	PPh ₃	88	9
7	NMP	1.2	P(<i>o</i> -tol) ₃	92	11
8	NMP	1.2	PtBu ₃	70	27

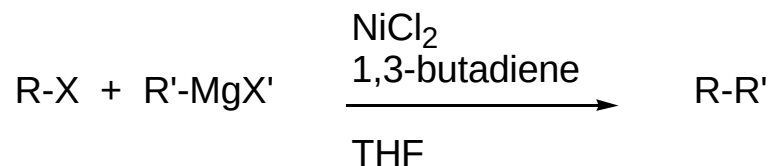
DMAc: *N,N*-Dimethylacetamide; NMP: *N*-methylpyrrolidinone

Frisch, A. C.; Shaikh, N.; Zaph, A.; Beller, M.* *Angew. Chem. Int. Ed.* **2002**, 41, 4056.

Kumada Coupling (sp^3-sp^3)

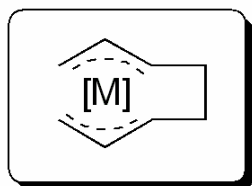
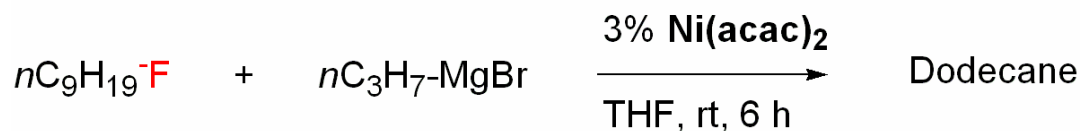
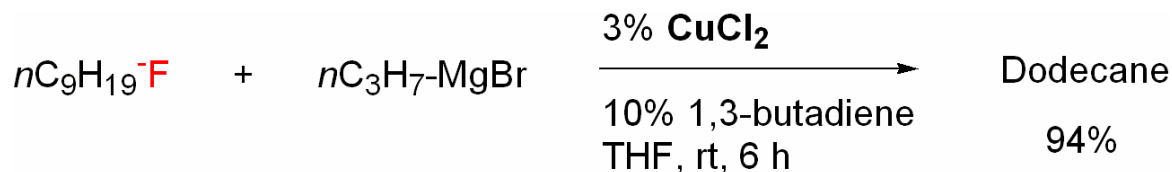


Kumada Coupling ($\text{sp}^3\text{-sp}^3$)

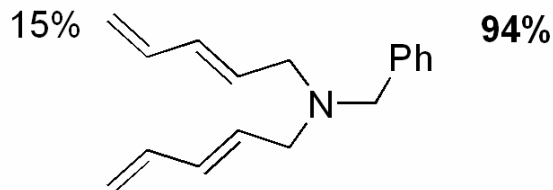


entry	R-X	R'-MgX'	Yield (%)
1	$n\text{C}_{10}\text{H}_{21}\text{-Br}$	$n\text{Bu-MgCl}$	100
2	$(4\text{-Br-C}_6\text{H}_4)\text{CH}_2\text{CH}_2\text{Br}$	$n\text{Bu-MgCl}$	100
3	$\text{PhCH}_2\text{CH}_2\text{OTs}$	Et-MgBr	87
4	$n\text{Bu-OTs}$	$n\text{Bu-MgCl}$	100
5	Et-OTs	Ph-MgBr	56
6	$n\text{Oct-Br}$	Ph-MgBr	90
7	$n\text{Oct-Br}$	$i\text{Pr-MgCl}$	72
8	$n\text{Oct-Cl}$	$n\text{Bu-MgCl}$	96

Kumada Coupling (Alkyl Fluorides)



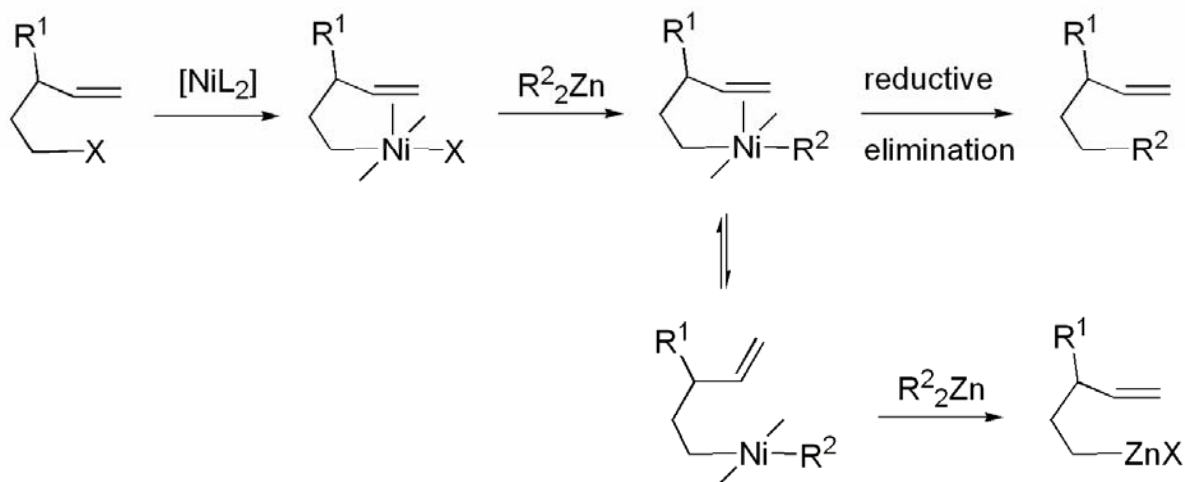
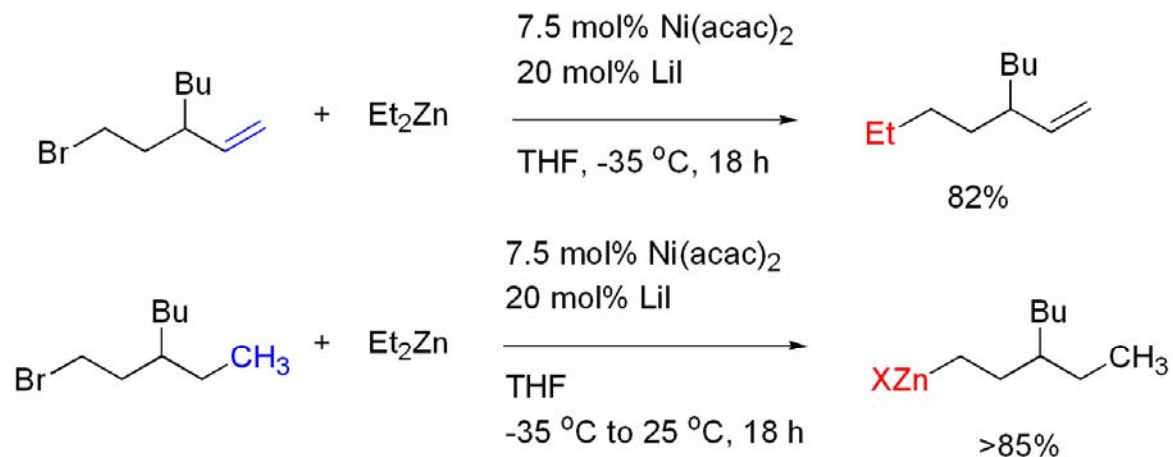
additive:	Yield
30% 1,3-butadiene	11%
100% 1,3-butadiene	58%
200% 1,3-butadiene	65%



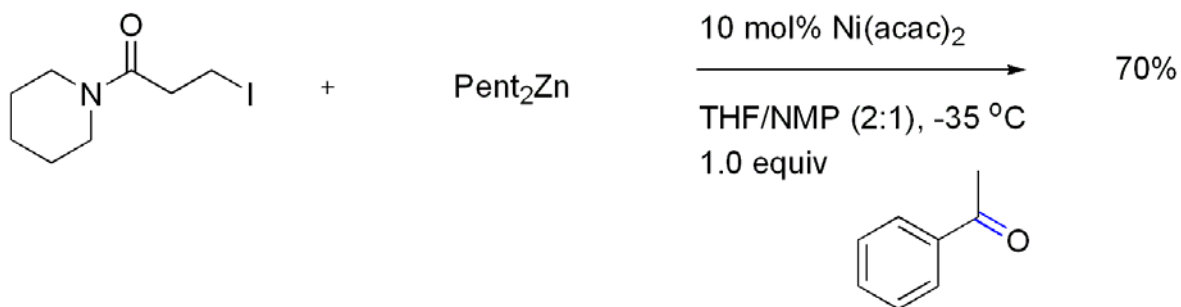
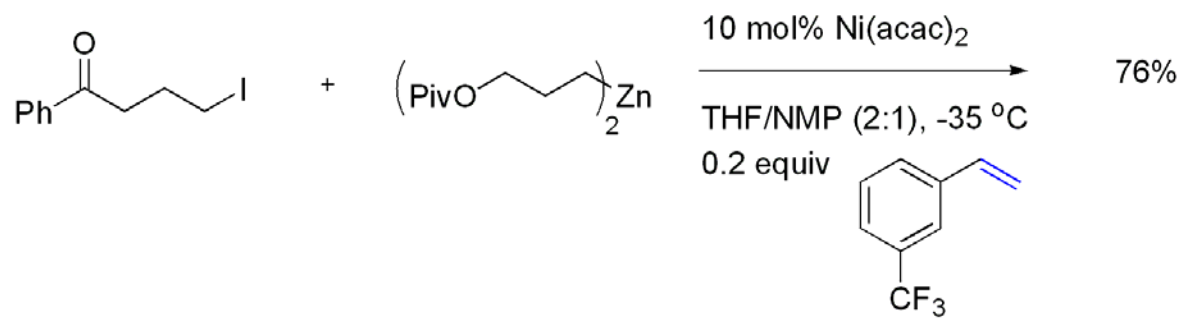
Terao, J.; Ikumi, A.; Kuniyasu, H.; Kambe, N.* *J. Am. Chem. Soc.* **2003**, 125, 5646.

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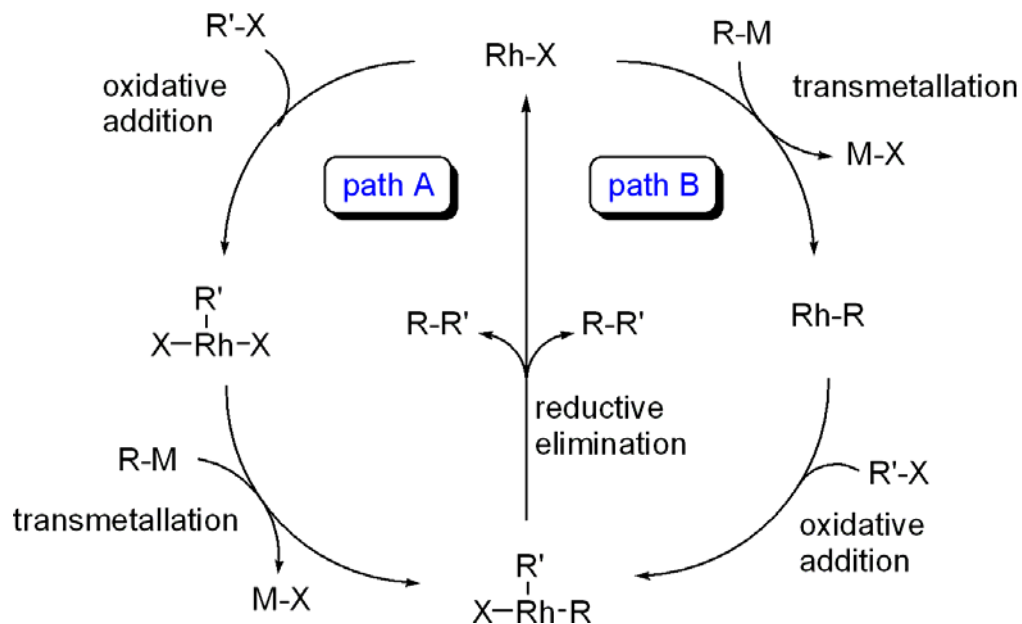
Negishi Coupling



Negishi Coupling

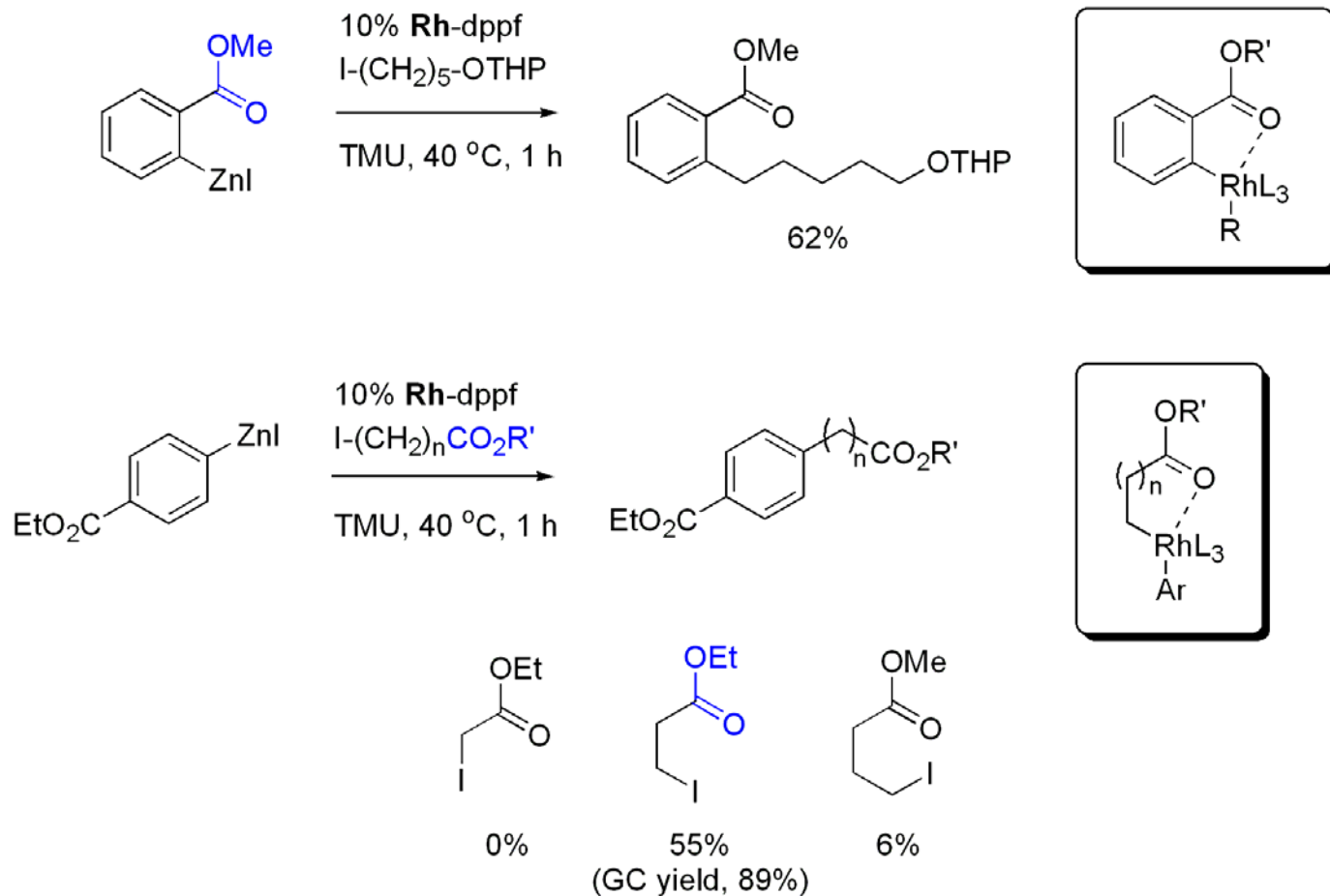


Rh in Negishi-Type Coupling



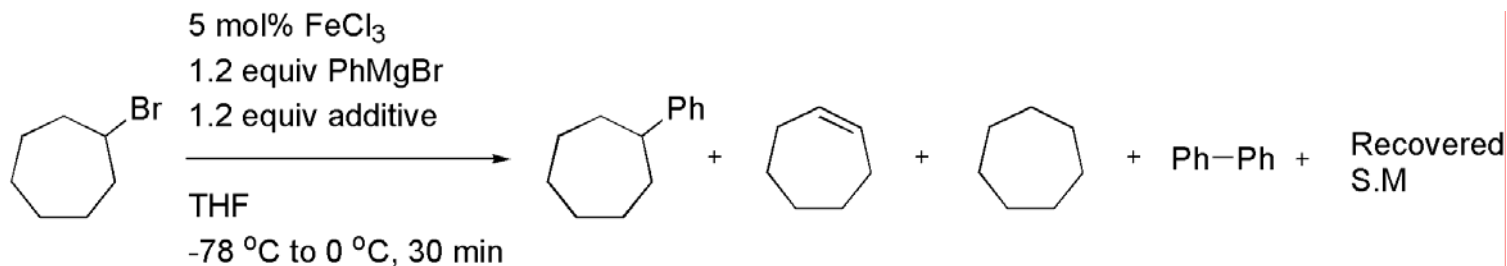
Takahashi, H.; Inagaki, S.; Nishihara, Y.; Shibata, T.; Takagi, K.*
Org. Lett. **2006**, 8, 3037.

Rh in Negishi-Type Coupling



Takahashi, H.; Inagaki, S.; Nishihara, Y.; Shibata, T.; Takagi, K.*
Org. Lett. **2006**, 8, 3037.

Fe in Kumada-Type Coupling

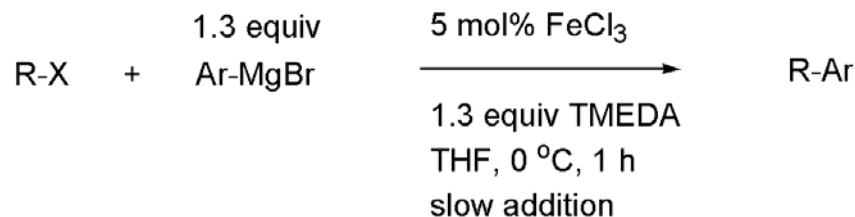


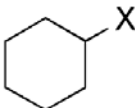
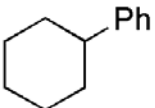
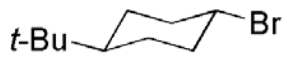
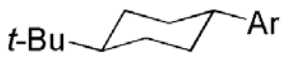
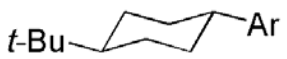
entry	additive					
1	none	5	79	0	6	4
2	Et ₃ N	3	78	0	5	11
3	<i>N</i> -methyl morpholine	8	72	0	5	4
4	DABCO	20	2	0	3	75
5	TMEDA	71	19	3	10	trace

TMEDA: *N,N,N',N'*-tetramethylethylenediamine

Nakamura, M.*; Matsuo, K.; Ito, S.; Nakamura, E.* *J. Am. Chem. Soc.* **2004**, 126, 3686.

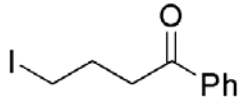
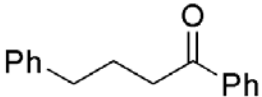
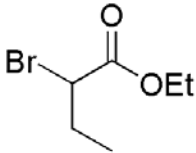
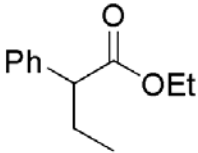
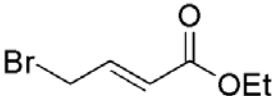
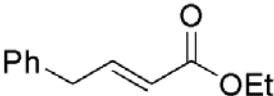
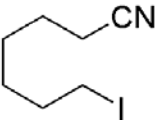
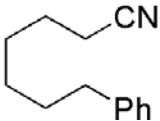
Fe in Kumada-Type Coupling



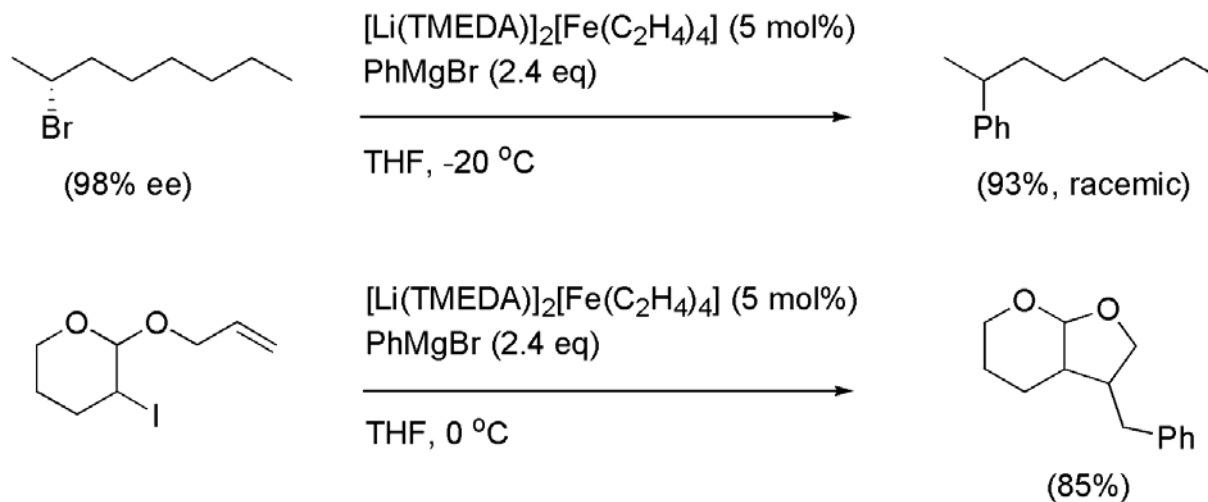
entry	alkyl halide	Ar-MgBr	Product	yield
1		Ph-MgBr		99% (X=I) 99% (X=Br) 99% (X=Cl)
2	$n\text{C}_8\text{H}_{17}\text{-X}$	Ph-MgBr	$n\text{C}_8\text{H}_{17}\text{-X}$	97% (X=I) 91% (X=Br) 45% (X=Cl)
3		4-MeOC ₆ H ₄ -MgBr		96%
4		4-MeOC ₆ H ₄ -MgBr		98%

Fe in Kumada-Type Coupling

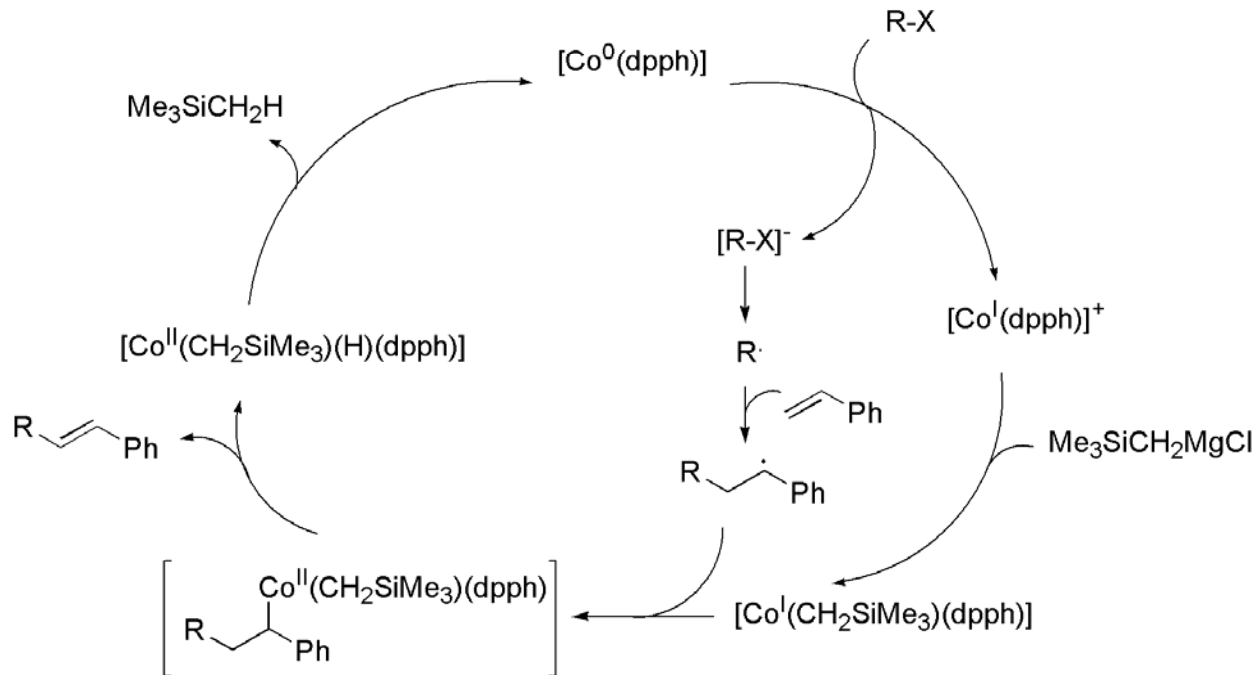
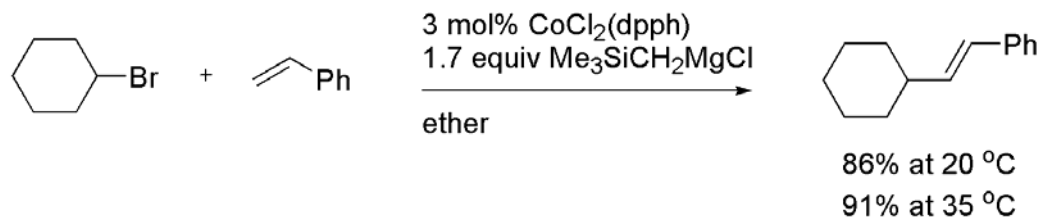
- ◆ Cross coupling reactions catalyzed by $[\text{Li}(\text{TMEDA})]_2[\text{Fe}(\text{C}_2\text{H}_4)_4]$

entry	substrate	RMgX	product	yield (%)
1		PhMgBr		91%
2		PhMgBr		87%
3		PhMgBr		94%
4		PhMgBr		83%

Fe in Kumada-Type Coupling



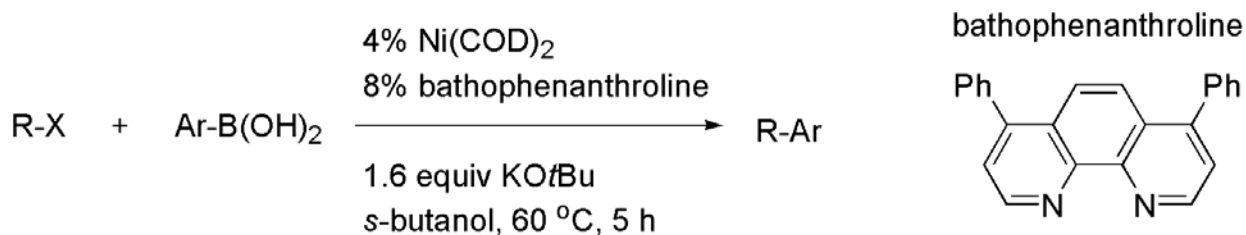
Heck-Type Reaction

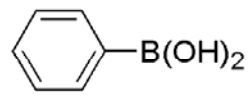
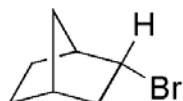
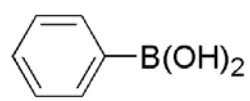
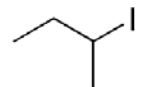
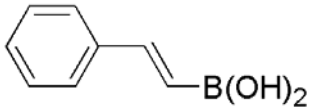
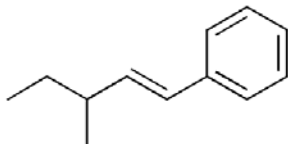


Ikeda, Y.; Nakamura, T.; Yorimitsu, H.; Oshima, K. *J. Am. Chem. Soc.* **2002**, *124*, 6514.
 Affo, W.; Ohmiya, H.; Fujioka, T.; Ikeda, Y.; Nakamura, T.; Yorimitsu, H.*; Oshima, K.*;
 Imamura, Y.; Mizuta, T.; Miyoshi, K. *J. Am. Chem. Soc.* **2006**, *128*, 8068.

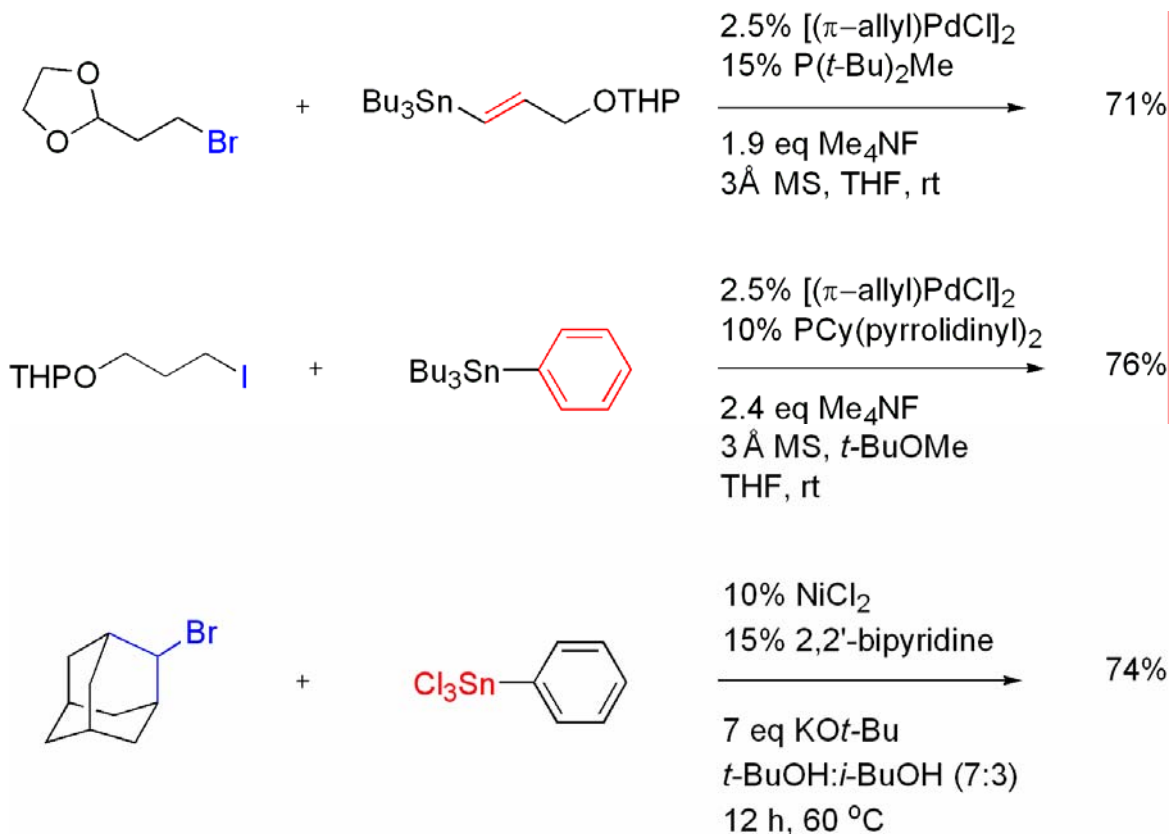
Suzuki Coupling

Secondary Alkyl Bromides and Iodides



entry	alkyl halide	boronic acid	product	yield
1				74%
2				71%
3				65%

Stille Coupling

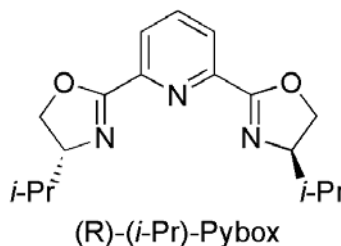
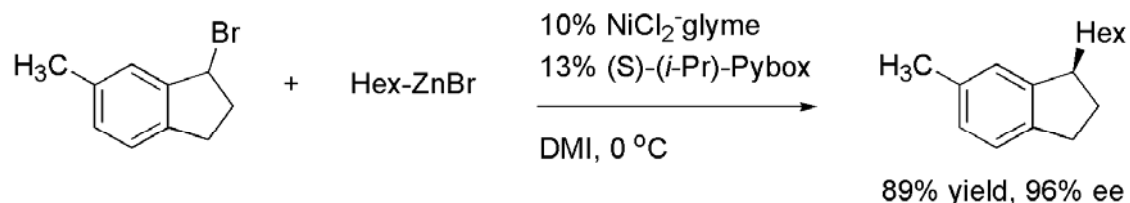
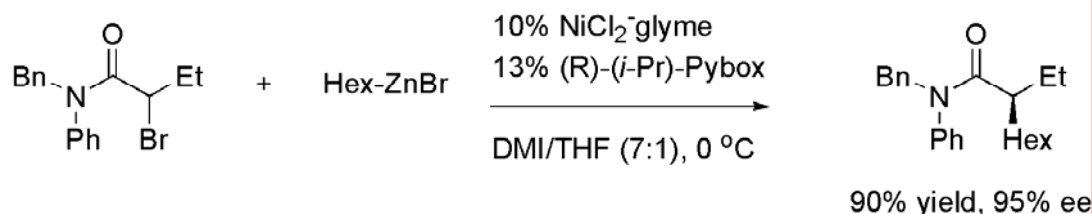
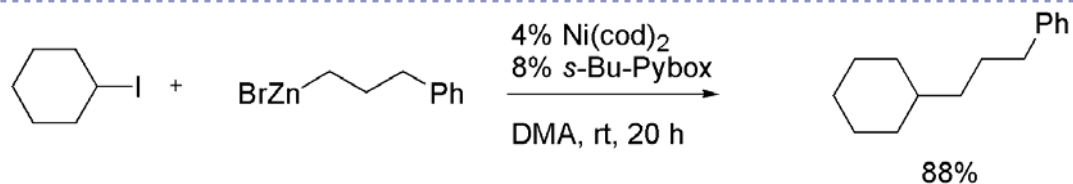


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Powell, D. A.; Maki, T.; Fu, G. C.* *J. Am. Chem. Soc.* **2005**, 127, 510.

Asymmetric Negishi Coupling Secondary Alkyl Bromides and Iodides



DMI: 1,3-dimethyl-2-imidazolidinone; DMA: *N,N*-dimethylacetamide

Zhou, J.; Fu, G. C. *J. Am. Chem. Soc.* **2003**, 125, 14726.

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Conclusions

- ◇ Cross Coupling reactions of unactivated alkyl halides have been successfully demonstrated.
- ◇ Expanding the scope of these processes
- ◇ Developing asymmetric reactions
- ◇ Understanding the mechanisms

References

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